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Smart NIR linear and nonlinear optical nanomaterials for cancer theranostics: Prospects in photomedicine

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ABSTRACT

Light-based diagnostics and therapy have become indispensable tools in the field of cancer nanomedicine. Various optical imaging modalities with tomographic capability have been developed to visualize cellular and organismic distributions of molecules. Microscopic pharmacokinetics and the tumor-targeting efficacy of nanoscale effectors can now be precisely evaluated. Moreover, phototherapy using intense laser light has been widely used for treating cancers. Using light-active nanoscale effectors, photothermal and photodynamic therapies on superficial tumors can be achieved with low-illumination lasers. Consequently, for the next generation of photo-medical techniques, the use of near infrared (NIR) excitation sources on NIR-activatable nanoparticles may offer deeper light penetration owing to less extensive scattering and absorption by endogenous chromophores in the NIR spectral region. Therefore, treatments and biodetection within higher tissue volumes and with less side effects (e.g. overheating) may be successfully implemented. This comprehensive review covers the state-of-the-art technologies on (a) advanced laser light sources appropriate for deep tissue theranostics, (b) types of laser interactions with pure-NIR and NIR-upconverting nanomaterials, (c) current development of NIR and multiphoton nanoparticles, (d) application fields of NIR nanomaterials in cancer theranostics, and (e) nanotoxicology of NIR nanoscale effectors for cancer treatment.

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Contents

1.	Introduction	90
2.	Contrast mechanisms of NIR nanomaterials	92
	2.1. Single-photon transition for fluorescence contrast agents	93
	2.2. Single photon phosphorescence	94
	2.3. Single photon Raman scattering	94
	2.4. Photoacoustic contrast	94
	2.5. Surface plasmon resonance effect	95
	2.6. Quantum confinement effects	95
	2.7. Rare-earth dopant based single photon up-converting process	95
	2.8. Multiphoton nonlinear optical processes	96
3.	Choice of light source and wavelength for NIR deep tissue theranostics	97
	3.1. Nd:YAG and Nd:YVO₄ lasers	99
	3.2. Cr:forsterite lasers	99
	3.3. Yb:fiber lasers	99
	3.4. Soliton-self-frequency shifted Er:fiber lasers	00
	3.5. Laser diodes	00
4.	Smart NIR linear and nonlinear optical nanomaterials	00
	4.1. NIR linear optical nanoparticles	00
	4.2. NIR nonlinear optical nanoparticles	01
	4.3. Multiphoton up-conversion of lanthanide nanoparticles	01
5.	Application fields of NIR nanomaterials in cancer theranostics	05
	5.1. Biosensing assays	05
	5.2. NIR imaging methods	06
	5.3. NIR fluorescence for image-guided surgery	07
	5.4. NIR photo-triggered drug release	08
	5.5. Photo-thermal therapy	08
	5.6. Upconversion induced photodynamic therapy	08
	5.7. Photo-dynamic therapy and photo-thermal ablation combined with NIR detection	10
6.	Nanotoxicology: concerns about the biosafety of NIR nanomaterials	12
	6.1. Methods to analyze the toxicology of NIR nanomaterials	12
	6.2. Toxicity of NIR nanomaterials	13
	6.3. Regulatory issues on nanomaterials	14
7.	Current challenges and future perspectives	19
	7.1. Sensitivity and background interference	19
	7.2. Technical hurdles and potential solutions	20
8.	Concluding remarks	21
	Acknowledgements	22
	References	22

1. Introduction

With the advance of cancer biology, the scientific community is gradually obtaining a better understanding of the heterogeneity of tumors, the pathophysiology of their growth, and the establishment and seeding of metastasis. New strategies using targeting therapies have been developed for personalized and effective treatment of tumors [1,2]. In recent years, with advances in nanotechnology, nanoparticles have been considered as effective vehicles for targeted drug delivery. With appropriate designed [3] size, surface chemistry or coating polymers, and conjugated antibodies, these "nanomedicines" can be engineered to circulate longer in blood, passively overcome biological barriers, actively target cancer cells, and penetrate into tumor tissues with high specificity. After targeting, cancer tomographic visualization (single or combined luminescence/magnetic resonance imaging (MRI)/positron emission tomography (PET) contrast agents) can be employed as well, and the release and penetration of drugs into tumors can be further triggered by the tumor chemical microenvironment (e.g., lower pH in tumors) or photon activation (e.g., photodynamic effects) of nanomaterials [4,5]. Moreover, photon- or magnetic field-activated hyperthermia is of great interest for cancer theranostics [6–8]. Therefore, the application of nanomedicines has become one of the clinically important and promising fields in cancer diagnosis and therapy. The *National Cancer Institute* at the *National Institute of Health* have recognized this and thus documented that nanotechnology offers an amazing, paradigm-shift opportunity to make significant advances in cancer diagnostics and therapy [9–11].

To evaluate the targeting efficacy, pharmacokinetics, and pharmacodynamics of nanomedicines *in vivo*, nanomaterials are usually designed to provide contrasts in various modalities of molecular imaging [12,13], so that they can be visualized and their time-course dynamics can be tracked on different scales. The location of multiple tumors can be identified and the treatment response can thus be followed up. PET uses gamma photons emitted from tagged radioactive tracers to map

nano-agents [14]. This whole body imaging modality has the best sensitivity and precision for mapping the bio-distribution of nanomaterials. However, PET-based imaging does not yield information on the structure of tissues and organs. Besides, the centimeter-scale spatial resolution is not sufficient for satisfactorily resolving sizes and precise locations. To put tumor imaging in a meaningful context, PET is usually combined with X-ray computed tomography (CT) which provides millimeter-scale resolution. Metallic nanoparticles can be used to enhance the contrast in CT imaging [15–17]. Using MRI, the resolution can be further improved to 100 µm, which enables detecting earlier stage tumors. Nanomaterials with chelated metallic ions [18], Gd³⁺ doped nanoparticles or superparamagnetic iron oxides [19] can enhance the T1 and T2 contrasts in MRI. Different modalities of images can be registered and combined together by using the same contrast agents with multiple detection mechanisms. One such proof of concept example is hexamodal porphyrin-phospholipid (PoP)-coated upconversion nanoparticles (UCNPs), which demonstrate NIR fluorescence; can be used for up-conversion imaging, photoacoustic imaging, PET and Cherenkov luminescence imaging with ⁶⁴Cu isotopes; and provide possibilities of CT and MRI contrast [20].

For understanding the whole-body distribution and metabolism of nanomedicines, cellular level imaging is required to monitor their pharmacokinetics and pharmacodynamics in detail. Compared with other molecular imaging modalities, optical imaging has the unique advantages of sub-cellular spatial resolution, high temporal resolution, the ability to employ versatile labeling tools, and sensitive detection of molecules at low concentration levels. Researchers have used various optical microscopy techniques to resolve the vascular permeation, diffusion, docking, and cellular internalization of nanomedicines in vivo. Except for the targeting function, the temporal progression of treatment-induced changes in the tumor microenvironment, recruitment of immune cells, and removal of cancer cells can also be evaluated. Under this observation scheme, nanomaterials need to be optically contrasting for being able to visualize their dynamics and cellular responses. Interestingly, as the size of these nanomaterials decreases down to several nanometers, their unique physical properties can yield prominent optical contrasting in a variety of optical microscopy modalities. Examples of these are surface plasmon resonance (SPR) absorption in metal nanoparticles [21], quantum confined effects of exciton fluorescence in semiconducting quantum dots [22], conductor to fluorescent semiconductor transition in gold nanodots [23–26], defect fluorescence in nanodiamonds [27,28], long living and anti-Stokes efficient up-converting nanoparticles, and surface-state luminescence of iron oxides [29]. These sensitive optical contrasts can also depend on the chemical microenvironment to reflect the cell physiological conditions such as pH values [30] and oxygen partial pressure pO_2 [31–35]. Besides, strong SPR absorption of metal nanoparticles can effectively convert light energy into heat and reactive oxygen species through photo-thermal and photodynamic effects, respectively [36]. These photo-physical or photo-chemical functions of nanomaterials can be excited on demand to damage cellular membranes or to locally increase the oxidative stress. The release and spread of therapeutic agents can thus be controlled and intentionally promoted. As a result, except for the role of a drug carrier, nanomaterials themselves can both serve as contrast agents in diagnostic molecular imaging and therapeutic means in photo-medicines [37]. Such integrated function of nanomaterials, often termed theranostics [38], can be used to optimize the efficacy of therapy to address patient-specific characteristics of cancer.

To achieve these merits of photo-nanomedicine for deep tissue diagnosis and treatment, NIR excitation sources for NIR-to-NIR-activated nanomaterials are being actively studied; this technique allows excitation in the NIR spectral range where light absorption and scattering from biological tissues are minimized, as well as an increased light penetration combined with locally induced laser hyperthermia. Illuminated at the so-called "biological window" or "water window" (~700–950 nm, 1000–1300 nm, and ~1600–1850 nm, corresponding to the 1st, 2nd, and 3rd optical windows, respectively), the NIR light minimally interacts with physiological constituents (i.e., pigments, proteins, coenzymes, water) and thus negligently affects the wavefront propagation. Such synergy of laser engineering and material sciences has broadened the horizon for the *in vivo* exploration of cancer biology, enabling deep tissue treatment of tumors. To achieve this goal, in the last decade many research efforts have been devoted to obtain nanomaterials capable of absorbing in the NIR range, with numerous biological applications. Nevertheless, longer excitation wavelengths degrade the spatial resolution of *in vivo* imaging. Higher excitation intensity may reshape the nanoparticles and degrade the NIR response properties [39]. Besides, for several functional nanomaterials the efficient excitation bands are in the visible range only [21–28]. To expand the advantages of NIR excitation of nanomaterials, these challenges need to be overcome with new chemical designs and chemical architectures in combination with new excitation schemes and detection/treatment platforms.

With advances in nonlinear optics and introduction of new imaging methods, NIR light can now be used to excite various optical contrasts of nano-agents to probe biological specimen both functionally and structurally with increasing spatial and temporal resolution. The nonlinear optical contrasts can be generated at a least invasive wavelength and power level. Simultaneously, significant interest driven by the field of nano-toxicology has recently emerged to understand how biological specimens respond to nano-engineered materials of various size and composition and with different surface properties. Besides, the visible absorption band of most functional nanomaterials becomes NIR-excitable through multiphoton processes allowing *in vitro* and *in vivo* multifunctional imaging and combined therapy using noble and magnetic nanoparticles [29,40]. The multiphoton yield of NIR excitation can be enhanced through the structural plasmon resonance [40] or surface modification [29]. Different NIR contrast mechanisms can thus be realized on single-type nanomaterials (e.g., gold, quantum dots, Fe₃O₄, polymers, carbon-based nanoparticles) to produce multiphoton emission, photoacoustic responses, and local hyperthermia. For example, NIR absorption and fluorescence were recently demonstrated for Fe₃O₄ nanoparticles [29], making these a versatile platform for photo-thermal therapy and two-photon fluorescent bio-imaging. The development of fluorescent Fe₃O₄ nanoparticles enables whole-scale cancer theranostics, combining optical microscopy with well-established MRI. Therefore, understanding the current development of these NIR-active nanoparticles combined with laser systems will



Fig. 1. Overview of NIR nanomaterials' applications for IMAGING- as contrast agents for X-ray, PET, MRI, fluorescence, multicolor visible-to-NIR luminescence, THERAPY by NIR-activated PDT therapy, chemo(drugs) release, hyperthermia combined with radionucleotide action, BIOSENSING, where different factors (temperature, chemicals, pH, metal ions) modulate the lifetime and intensity ratio of luminescence, and TARGETING on desired tissue sites.

inspire researchers to exploit "linear and nonlinear optical nanomaterials" beyond the conventional lanthanide-based upconverting nanoparticles.

Indeed, development of NIR photo-functional nanoparticles has promised new strategies to visualize tumor microenvironments *in vivo* as well as the improvement of a remote triggering of photodynamic or photo-thermal therapy and photo-induced chemical bond dissociation. For *in vivo* pre-clinical studies, these NIR multiphoton contrast agents provide a better anatomical resolution and detection sensitivity than conventional clinical MRI or X-ray CT.

Consequently, the core focus of this review article is to provide a concise overview of newly developed NIR-to-NIR nanomaterials and their potential applications in cancer targeting and deep-tissue imaging. This review goes well beyond the state-of-the-art technologies, by unifying the latest advances in designing smart NIR-to-NIR multiphoton nanotechnology for cancer theranostics. These newly developed multifunctional NIR-to-NIR nanomaterials and unique optical imaging methods will certainly enhance the capabilities of tumor targeted imaging and hold promise in cancer diagnostics and therapeutics, tackling crucial biomedical questions (Fig. 1).

2. Contrast mechanisms of NIR nanomaterials

Depending on the light-matter interaction pathways, the contrast characteristics of NIR optical imaging may arise from linear or nonlinear optical processes. Linear optical processes include absorption, scattering, interference, fluorescence, phosphorescence, and Raman scattering (Fig. 2). Absorbed light may be further converted into heat or ballistic acoustic waves.

Nonlinear optical processes are important contrast mechanisms for realizing NIR imaging. These processes are excited/ activated by intense laser pulses and include multiphoton absorption and fluorescence, second harmonic generation (SHG), third harmonic generation (THG), coherent anti-Stokes Raman scattering (CARS), stimulated Raman scattering (SRS), and multiphoton up-conversion luminescence (Fig. 3). These contrasts either can yield morphological information or report the molecular distribution to reveal the cancer biology and pharmacokinetics of nanomedicines *in vivo*. Different contrasts have their own *pros* and *cons* in different contexts. In the following sections, we will introduce the major types of linear and nonlinear optical contrasts of nanomaterials, and their corresponding imaging modalities.



Fig. 2. Types of single-photon contrast mechanisms.



Fig. 3. Schematic of the energy diagram of the absorption and emission features of nanomaterials based on the single-photon and multiphoton transitions.

2.1. Single-photon transition for fluorescence contrast agents

Fluorescence processes involve only transitions of electrons within molecules, and their excitation and emission bands are determined by molecular composition and structures. They can be single-photon excited at low intensities (Fig. 2). Light excitation of the absorption band of fluorophores results in the excitation of ground-state electrons. Through a fast (\sim 300 fs) internal conversion or vibrational relaxation [41], thermalized electrons accumulate in *quasi*-steady states and then relax back to their ground states by releasing photons. Typically, this fluorescence relaxation process occurs on the nanosecond (singlet-singlet fluorescence) or micro to millisecond (triplet-singlet phosphorescence) scale. The red-shift of the emission wavelength allows the suppression of excitation photons with filters and the enhancement of detection sensitivity with photomultiplier tubes. Therefore, among all optical contrast processes, fluorescence is the most sensitive and specific one (Fig. 3) for biomedical imaging. To achieve NIR fluorescence in nanomaterials, a straightforward strategy is to cross-link or anchor NIR dyes such as cyanine molecules with or to the surface of nanoparticles. To reduce the complexity of synthesis and to avoid the concern of extra toxicity, inherent fluorescence of semiconductor quantum dots (e.g., CdSe [22], Ag₂S [42], CulnS₂ [43]) or carbon nanotubes [44–46] has been developed. Exploiting type-II band-alignment between core-shell semiconductors, NIR fluorescence can be further tuned to longer wavelengths [47]. For indirect bandgap materials such as silicon, their nanometer sizes increase the momentum uncertainties of electrons and assists in the indirect relaxation of electrons. Silicon nanocrystals exploit these properties to obtain more efficient NIR fluorescence [48,49]. Most of other luminescent nanomaterials rely on defect states or localized states to yield less efficient and relatively broad NIR emission bands. This luminescence becomes prominent for nanomaterials with large surface-to-volume ratio, such as carbon nanodots [50–54], nanodiamonds [27,28], and gold clusters [55]. Using these fluorescence labels, injected nanomedicines can be visualized and tracked using a bright-field fluorescence imaging system. Nanomaterials may be illuminated at a specific excitation wavelength and the diffused fluorescent photons collected and form a whole body image on a sensitive charge-coupled device (CCD) camera. Such an *in vivo* imaging system can be used for evaluating the whole body circulation and the targeting of nanomedicines. Just like a conventional camera system, the angular resolution of this imaging modality is typically determined by the emission color of labels, the aperture of cameras, and the pixel densities of CCD CMOS chips.

To take a closer look at microscopic scales *in vivo*, the use of laser scanned confocal microscopy techniques is required. At each excitation point, the out-of-focus fluorescence will be rejected by a confocal pinhole before photomultiplier tubes and the tomographic distribution of nanomedicines *in vivo* can be revealed in a three-dimensional (3D) image stack with submicron resolution. In this case, the imaging depth will be limited by the scattering properties of tissues, and is typically 100–150 µm for turbid tissues such as skin.

In addition to the fluorescence intensity, the fluorescence lifetime is another contrast mechanism of fluorescence that is often highly sensitive to the molecular microenvironment. Related imaging modalities include fluorescence resonant energy transfer (FRET) microscopy, fluorescence lifetime imaging microscopy (FLIM), and time-gated fluorescence microscopy. The FRET and FLIM modalities can be used to evaluate microscopic drug release from theranostic nanoparticles. Before drug release, FRET events occur between the donor (nanoparticles) and acceptor (fluorescent drug). After release, the acceptor diffuses away and only the donor fluorescence can be detected. This evaluation scheme has been realized for quantum dots [56,57], polymer-based micelles [58], and mesoporous silica nanoparticles [59]. Differences in the fluorescence lifetimes across molecules can be used to differentiate the targeted molecules from other molecules [60,61] (e.g., bound and unbound reduced nicotinamide adenine dinucleotide (NADH) fluorescence) or from a non-specific background [62]. For instance, the auto-fluorescence lifetime of endogenous chromophores is typically ~1 ns. By employing nanomaterials with fluorescence lifetimes on the order of several tens of nanoseconds (e.g., quantum dots (QDs) and nanodiamonds) or luminescence lifetimes on the order of micro- to milliseconds (lanthanide doped nanoparticles), and by performing time-delayed and gated integration of photons after the excitation pulse, it is possible to realize a background-free and thus possibly ultrasensitive detection and fluorescence imaging of nanomaterials. This concept has been recently applied to fluorescent diamond nanoparticles for tracking diamond labeled cells in an environment with large auto-fluorescence [62].

2.2. Single photon phosphorescence

For some photosensitizing molecules, such as porphyrins, photo-excited electrons can couple to triplet states through intersystem crossing. Photon emission of phosphorescence occurs when triplet-state electrons relax back to singlet ground states, whose wavelengths typically range from red to NIR (Fig. 3). In contrast to fluorescence, this forbidden transition process of electrons results in a microsecond to millisecond lifetime after excitation. The lifetime is so long that phosphorescence could be quenched by collisions of triplet oxygens in water. This quenching is typically responsible for dimming the phosphorescence intensity and shortening the phosphorescence lifetime. Exploiting this property, the PO_2 levels in vasculature and tissue can be quantitatively evaluated based on changes in the phosphorescence intensities or lifetimes. However, these photosensitizers are toxic to cells and not stable in biological environment. Nanomaterials, in this case, can host these phosphorescence probes to study PO_2 in the eye retinae [63], in the cerebral vasculature and tissue [64], and in the niche of hematopoietic stem cells [65]. Microscopic PO_2 imaging can be performed with three-dimensional (3D) sub-micron resolution, which is useful for investigating hypoxic microenvironments in wound healing, tumor growth, and stem cell niches.

2.3. Single photon Raman scattering

Different from elastic scattering (Fig. 2), in the Raman scattering process, the incident light (photons) interacts with the vibrations (phonons) of the target molecules and the scattered photons may have lower (the Stokes line) or higher (the anti-Stokes line) energy. Each molecule, owing to its characteristic bonding structure and vibrational modes, has its own spontaneous Raman spectrum. Depending on the size of molecules and the strength of bonds, the vibration frequency of interest can range from very low (5 cm⁻¹) to 4000 cm⁻¹. Strong Raman peaks can serve as contrasts for microscopic molecular imaging. For biological samples, NIR-excited Raman spectra can reduce the interference from auto-fluorescence. However, the Raman scattering intensity is proportional to $1/\lambda^4$, which means that the signal is weakened by an order of magnitude for wavelengths in the NIR range. Therefore, there is a trade-off on selecting a proper wavelength for excitations associated with Raman scattering.

2.4. Photoacoustic contrast

Traditionally, staining dyes have been used in bright-field microscopy for absorption contrast of cells. Cells with intrinsic pigments, such as red blood cells, can be easily observed *in vivo* without any staining [66]. Using photoacoustic imaging, the light absorption characteristics of materials can be exploited to convert light energy into acoustic waves for background-free and deep-tissue tomography. The main advantage of the photoacoustic contrast method is its flexibility: it can either use optical excitation or acoustic detection as the aperture function for determining the imaging resolution [67]. In the ballistic excitation regime, where the imaging depth is smaller than the inverse of the photon scattering constant, a focused Gaussian beam can still maintain a point spread function (PSF) of sufficient quality for the microscopic scale. Beyond the ballistic regime, the optical wavefront gradually loses its coherence and the PSF can easily increase beyond 100 µm. In this case, less divergent acoustic waves have better resolution using phase-array detection. This flexibility of photo-acoustic imaging extends the microscopic scale imaging to a depth much deeper than what can be achieved using optical coherence tomography (OCT). A widely exploited photo-acoustic contrast is the absorption of hemoglobin in red blood cells. Fine blood vasculature can be mapped in a larger field of view at a deeper imaging depth [67]. A functional photoacoustic microscope was developed, which provides multi-wavelength imaging of optical absorption and allows high spatial resolution beyond this depth limit with a ratio of maximal imaging depth to depth resolution greater than 100. This capability is useful for studying angiogenesis in tumor microenvironments.

Among many nanomaterial systems, noble metal nanoparticles are ideal NIR photoacoustic contrast agents for *in vivo* nanoscale molecular imaging. Recently, Bao et al. reported the use of gold nanoprisms as novel contrast agents for the hybrid technique of optoacoustic imaging in mice gastrointestinal tumors [68,69]. They have demonstrated a huge absorption crosssection of such nanoparticles owing to their free electrons. The SPR effect plays a major role in increasing the absorption cross-section and manipulating peak absorption wavelengths.

2.5. Surface plasmon resonance effect

Surface plasmons are dipolar excitations related to the density waves of free electrons in metals. For isolated nanoparticles, surface plasmon modes are standing waves with zero momentum and can thus easily interact with photons [21,70]. According to Mie theory, this dipolar absorption makes the dielectric constants of a metal negative in a certain range of wavelengths, which can locally enhance the electric field on a spatial scale much smaller than that of the excitation wavelength [71]. Typically, for spherical solid metal nanoparticles, the SPR wavelengths are in the visible range. To achieve SPR in the NIR wavelength range, the geometry of metal nanoparticles needs to be tailored into nanorods, nano-shells, or triangular nanoplates. Another strategy is to employ materials with lower free-carrier densities, such as conducting metal oxides or doped semiconducting nanocrystals [72]. This SPR-enhanced electric field can significantly increase the yield of originally weak signals such as those in Raman scattering [73], harmonic generation [40,74–76], and multiphoton fluorescence [40]. This signal enhancement feature of SPR can be employed for detecting trace amounts of molecules. The SPR effects can also enhance the Raman scattering in nanomaterials, especially for the enhancement of NIR excited ones. For example, using micro-Raman spectroscopy, the detection of individual molecular vibration signals became possible once the surfaceenhanced Raman scattering (SERS) (Fig. 2) was involved after combining long-range electromagnetic and short-range chemical enhancements [73,77]. Currently, molecular tags on Au nanomaterials have become a powerful tool for developing successful *in vivo* disease site tracking methods [78], sensing the intracellular pH environment [79], and screening circulating tumor cells [80] by using NIR lasers.

2.6. Quantum confinement effects

In quantum mechanics, the particle-in-a-box model explains how free particles can occupy certain positive energy levels when they are confined to a low-dimensional potential well. In a nanometer-scale semiconductor quantum well, there are quantized eigenstates of electron wave functions in the conduction and valence bands. For narrower well widths, the transition bandgap is larger, yielding a fluorescence blue-shift. For low-dimensional nanomaterials, 3D confinement of electron or exciton wave functions yields similar effects. For instance, smaller sizes of CdSe semiconductor quantum dots have shorter exciton absorption peaks and emission wavelengths [81], which can be tuned from blue (~480 nm) to red (~650 nm). This color tunability can be used for multiplexed optical coding of biomolecules [82]. For indirect-bandgap silicon nanocrystals, this confinement will also increase the transition gaps of free excitons and localized states [83]. The tuning range is wider from 400 nm to the NIR 750 nm wavelength. Another type of quantum confinement is defect-related localized states. Depending on the domain environment and size, different transition sites may have different excitation and emission peaks. Fluorescent carbon nanodots have this feature, and the emission wavelength can be tuned via excitation ranging from 400 nm to the 700-nm wavelength of NIR [50–54].

2.7. Rare-earth dopant based single photon up-converting process

Up-conversion of photons is a sequential absorption of two or more photons (typically NIR photons) by materials that leads to the emission of light at a shorter wavelength than the excitation one (the so-called anti-Stokes emission) [84]. This

phenomenon usually occurs in solid-state materials doped with d-block and f-block elements. The interaction between doping ions and lattices forms ladder-like arrangement of energy levels with similar spacing, and allows sequential excitation of electrons to an even higher excited state. The up-conversion processes can roughly be divided into three categories: excited state absorption (ESA), energy transfer up-conversion (ETU), and photo avalanche (PA). For example, Yb³⁺ sensitized Er³⁺ or Tm³⁺ systems are usually co-doped in NaYF₄ and yield an efficient ETU [85]. The Yb³⁺ sensitized Ho³⁺ co-doped oxyfluoride glass ceramics yield a serial ESA and PA up-conversion [86]. These nonlinear optical processes should be distinguished from coherent up-conversion processes such as multi-photon absorption and harmonic generation. The major difference is that the photon up-conversion can be realized at low excitation intensities, while the coherent up-conversion cannot, which is owing to the engagement of real or virtual energy levels, respectively. For biomedical imaging, this unique optical contrasting mechanism has been used in rare-earth ion doped nanoparticles [85,87]. Multicolor and narrowband absorption/emission are achieved as well as color tunability [88] at the synthesis stage, perfect photo-stability of luminescence and both NIRto-vis and NIR-to-NIR emission can be obtained with these materials. However, for nanomaterials with a high surface-tovolume ratio, the excited electron clouds may easily couple with surface states and severely quench the up-conversion process. In this situation, an undoped shell coating on the up-converting core is required to keep the excited electrons away from fast multiphotons and non-radiative quenching by the surface ligands and water molecules [87].

Compared with organic dyes and quantum dots, up-conversion luminescence of rare-earth doped nanoparticles allows the up-conversion processes based on Yb³⁺ sensitized Er³⁺ or Tm³⁺ systems [85]. Owing to the common excitation at 980 nm, a NaYF₄ host combined with Yb³⁺/Er³⁺/Tm³⁺ guests permits deep-tissue excitation and avoids the excitation of auto-fluorescence owing to the tissue background, which is a severe drawback of most *in vitro* or *in vivo* optical Stokes ($\lambda_{exc} < \lambda_{emi}$) microscopy systems. However, the up-conversion process decays in typically several tens of microseconds to a millisecond [89], which sets a lower limit on the pixel dwell time and thus slows down the imaging speed of laserscanned confocal microscopy. For example, for a 512 × 512 pixels image, a pixel dwell time of 10 µs will result in the frame time of 2.62 s, which is too slow for imaging some kinetic phenomena and characteristic of biological systems. Therefore, in most cases, up-conversion contrast labels are applied to whole body wide field imaging and bright-field fluorescence microscopy [90].

2.8. Multiphoton nonlinear optical processes

According to the electromagnetism theory, the polarization density *P* of a material can be induced by light illumination. For low-intensity illuminations, the amplitude of *P* is linear in the applied electric field *E*. With the invention of ultrafast lasers, coherent light energy can be instantaneously delivered using picosecond (10^{-12} s) or femtosecond (10^{-15} s) pulses; as a result, the instantaneous intensity can reach unprecedented levels of 10^{12} W/cm^2 even for low power. Such strong coherent electric fields can drive electrons in materials away from atomic harmonic potentials, and anharmonic vibrations induce nonlinear polarization in materials. From the viewpoint of light-matter interaction, this nonlinear polarization involves multiphoton processes in which new photons with shorter wavelengths are generated. Because the signal yield depends nonlinearly on the intensity of light excitation, detectable nonlinear optical signals can only be generated within the confocal range around the focus. This feature of nonlinear optical signals defines a sectioning plane of images. Therefore, without using confocal pinholes, these modalities have intrinsic capability of sectioning microscopy *in vivo*. Besides, the effective PSF will be improved by $1/\sqrt{N}$, where N is the order of nonlinearity [91].

The SHG is an energy-conserved coherent generation of photons with doubled light frequency. The generation of nonlinear polarization does not require real electronic states. Therefore, NIR femtosecond lasers can generate SHG without any limitation on the excitation wavelength. Nonlinear polarization occurs in materials with non-central symmetry. Typically, homogeneous tissues or cells would not generate SHG signals. These signals can be generated by structured proteins such as collagens and muscles [92–96], spindle fibers [92–95], micro-tubulins [92–95], fibrous astroglial filaments around axons [97–99] and *zona pellucida* of mammalian oocytes [100]. The strong SHG of collagen has been used in optical virtual skin biopsy [101,102], in the context of dermatitis [103], aging [104], and tumor diagnosis [105]. Since most nanoparticles have a non-central symmetric crystal structure, they more or less carry permanent dipoles that can assist in the generation of SHG. Even for central symmetric metal nanoparticles, quadrupolar SHG can still be generated at the surface where symmetry breaks [106]. As the surface-to-volume ratio increases, non-symmetric shapes may also induce dipolar SHG [107]. Exploiting the SPR effects or resonance enhancement effects, weak SHG in nanoparticles can further be enhanced [40]. These SHG nanoprobes can be used as non-bleachable and non-blinking contrast agents for long-term cell tracking [108,109].

The frequency tripled THG process does not have symmetry restrictions. However, owing to the *Gouy* phase shift of focused Gaussian beams, the generation of THG before focusing will be cancelled by the one that is generated after focusing. This implies that homogeneous samples cannot generate THG. THG generation only occurs at interfaces between media with significantly different refractive indices [110] or on nanoparticles with sizes much smaller than the focal volume [74,75]. Similar to the differential interference contrast microscopy, the THG contrast can yield the *in vivo* morphology with a 3D submicron resolution. This contrast has been used for imaging cellular membranes, lipids [111,112], elastic fibers [113], hemoglobin [114], melanin [105], and granules in leukocytes [115,116]. To obtain strong THG signals in the visible wavelength (390–700 nm), where the microscope is designed to have high transmission, the excitation source must be an NIR laser with the wavelength longer than 1170 nm [117]. This will require special gain media and design of laser sources. For nanoparticles, the THG intensity will be proportional to their volumes [118,119]. Higher refractive index contrasts, such

as Si nanowires [120], would result in a larger THG yield. Similar to SHG, the yield of THG can also be enhanced by SPR effects [40,74–76].

Different from harmonic generation microscopy, Raman scattering-based CARS and SRS microscopy achieve contrast through molecular vibrations and thus, have more molecular specificity. In CARS microscopy, two photons from a picosecond pump (~850 nm) and one from the Stokes field (1.12–1.17 μ m) in the infrared range can coherently generate the anti-Stokes field in the red region (650–700 nm) [121,122]. This three-photon process involves a third-order nonlinear optical effect. To remove the non-resonant background associated with four-wave mixing and to improve the spectral resolution, SRS microscopy was developed [123]. Originally, only the C-H stretching mode and water were considered to exhibit sufficiently strong CARS or SRS responses for quick imaging frame rates. At present, video-rate SRS microscopy has been realized by improving the collection efficiency [124]. The commonly employed vibration modes of nanoparticles in CARS or SRS microscopy are the C-H stretching modes of polystyrene beads [121,122]. Other vibrational modes and corresponding contrast agents include the G mode vibration of carbon-nanotubes [125] and the C-C sp3 vibration of nanodiamonds [126]. These CARS contrasts of nanoparticles allow label-free long-term cell tracking for *in vitro* or *in vivo* studies [127].

Among all nonlinear optical contrasts, multiphoton fluorescence is based on intra-molecular electronic transitions and has the best molecular specificity. The CARS or SRS contrasts may yield information on molecular vibrations, but many vibration modes are commonly shared by different proteins or lipids. The theory of two photon fluorescence (TPF) was first proposed by Maria Göppert-Mayer in 1931. After the invention of femtosecond lasers, the laser scanned TPF microscopy was realized by Webb's group in 1990 [128]. In the TPF process, the electrons can be excited with photon energy slightly above half of the bandgap. Visible fluorescence can thus be instantaneously generated with infrared light sources. Just like SHG and THG microscopies, the TPF microscopy has depth discrimination owing to the nonlinear dependence of yields on the excitation intensity. Compared with single-photon fluorescence microscopy, this fluorescence excitation scheme has the benefit of reduced out-of-focus photo-bleaching on dyes, deep tissue NIR excitation, resolution improvement through nonlinear optical processes, and selective excitation of the dyes of interest. With the invention of genetic labeling with fluorescence proteins, TPF microscopy has become widely used in longitudinal studies in molecular cell biology in vivo [129]. The TPF efficiency of molecules or nanomaterials is evaluated by the two-photon action cross-section, which is the product of fluorescent quantum yield $\varphi_{\rm F}$ and the absolute two photon absorption cross-section σ_{2p} (GM). The TPF action cross-sections of endogenous fluorophores such as NADH are fairly low ($<10^{-4}$ GM) [129]. Most commonly used dyes and fluorescence proteins have 1–300 GM action cross-sections [129]. Using CdSe-ZnS quantum dots, this action cross-section can be increased to 50,000 GM [129], which allows a lower excitation dosage of light or greater imaging depth.

3. Choice of light source and wavelength for NIR deep tissue theranostics

Considering *in vivo* deep tissue imaging, visible light excitation has poor PSF performance owing to severe Rayleigh and Mie scattering from pigments and randomly oriented collagen networks. The absorption of melanin and hemoglobin (Hb) by the vasculature can further attenuate the contrasts in the 400–800 nm range of wavelengths, and strong excitation may induce photo-thermal damage. Two-photon excitation at wavelengths around or below 800 nm can also excite endogenous photosensitizers, such as porphyrins, and can generate reactive oxygen species (ROS) [130]. On the other hand, for wavelengths longer than 1300 nm, water absorption becomes the major limiting factor. High illumination intensity for deeptissue imaging also induces photo-thermal damage. Therefore, for most biological tissues, the penetration window for optical imaging is 800–1300 nm (Fig. 4A) [131,132]. The range of 1000–1300 nm would yield a better performance of imaging depth and less photo-damage associated with measurements. For tissues with weaker presence of pigments and collagen (e.g., brains), wavelengths in the 1600–1850 nm range can also be considered [133].

Considering the 1000–1300 nm penetration window of biological tissues, lanthanide ion doped up-conversion nanomaterials have become popular for use in NIR deep-tissue theranostics. The optical properties of lanthanide ions have been known for many years, and lanthanide doped crystals, glasses, and fibers have been used as optically active materials for compact solid state laser crystals [134–136], fiber lasers [137], TV/lamp phosphors [138], and IR quantum cutters [139]. The interest in lanthanide up-conversion has resulted in new discoveries, i.e., many up-conversion mechanisms have been discovered in lanthanide ions, such as ESA, co-operative energy transfer (CET), PA, or most ETU [140]. These discoveries opened up a new field for applications in volumetric displays [141], remote temperature fiber sensors [142], and upconversion lasers [143]. Recently, owing to chemical engineering of active-core-active-shell nanoparticles, novel routes of ETU have been designed such as energy migration mediated energy up-conversion [144–146], which are of special importance for biomedical applications.

There are three types of transitions in lanthanides (Fig. 4B), which are of some importance for biomedical applications:

(i) **Stokes visible emission under ultraviolet (UV) excitation** (typically <400 nm excitation for Eu^{3+} (~630 nm), Tb^{3+} (540 nm), Sm^{3+} (650 nm), and Dy^{3+} (570 nm) complexes). The complexes of Eu^{3+} and Tb^{3+} ions have been often used in bioassays or for bio-imaging, but require short-wavelength excitation. The spectral overlap of excitation bands or emission bands of these compounds with respective absorption or emission of endogenous chromophores decreases the sensitivity of bioassays or decreases the contrast of *in vivo* imaging. However, owing to the very long luminescence lifetimes of Tb^{3+} and Eu^{3+} (millisecond scale), time-gated techniques are efficient in removing background signals as well as for studying biochemical processes using the FRET technique [147].



Fig. 4. (A) Absorption (μ a) and reduced scattering (μ s') coefficients of major tissue components and representative tissues, Fluorescence spectra of major tissue components [NADH, DNA, elastin, collagen, flavin adenine dinucleotide (FAD)] combined with representative fluorescence spectra of commercial quantum dots and organic fluorophores, and reduced scattering coefficient of skin, brain and breast tissues (B), (stokes and anti-Stokes) luminescence spectra of lanthanide doped nanoparticles (C), sensitivity of available photo-detectors (D) and available discreet (horizontal red lines) and tunable (vertical lines indicate tuning range) state-of-the-art lasers. R5983, R928, and R3896 are part numbers of the Hamamatsu photomultiplier tubes.

(ii) Stokes emission in NIR (λ > 1 µm). Most lanthanides demonstrate Stokes emission with obviously higher quantum efficiency than the up-conversion processes. Some of these ions can generate emission in the NIR I or NIR II optical windows. The most prominent rare-earth ions for *in vivo* NIR imaging are Nd³⁺ (emission at 860 nm, 1060 nm, and 1330 nm), Er³⁺ (emission at 1530 nm) or Ho³⁺ (emission at 1450 nm) [148]. Most of the existing studies concentrated on LaF₃ [149,150], NdF₃ [151], and Y₂O₃:Yb/Er [152], but the most promising material is NaYF₄, e.g., NaGdF₄:Nd³⁺@

 $NaGdF_4$ and $NaYF_4$: Yb/Er/Ho/Tm/Pr@NaYF_4 core-shell down-converting nanoparticles [153,154]. Unlike LaF_3, fluorides of the NaYF_4 type are synthesized in a more predictable manner, i.e., the synthesis protocols, mono-distribution of size, bio-functionalization protocols, and ability to make core-shell designs are much better controlled and reproducible.

(iii) Anti-Stokes (up-conversion) emission under NIR photoexcitation – typically 980 nm excitation for Yb³⁺ sensitizers or 808 nm photoexcitation for Nd³⁺ sensitizers is used to achieve visible and multicolor emission from activators such as Tm³⁺, Er³⁺, Ho³⁺, and Tb³⁺. No organic chromophores absorb at 980 nm (absorption of Yb sensitizers) or 800 nm (absorption of Nd³⁺ sensitizers); thus, the signal to noise ratio is usually very high. Only water molecules exhibit an absorption band at ~980 nm, which under high photoexcitation densities may induce local overheating. These effects can be however diminished by using Nd³⁺ primary sensitizers, whose absorption cross-section at 800 nm is ~5 times higher than that of Yb³⁺ at 980 nm; in contrast, the absorption coefficient of water at 800 nm is ~20 times lower than that at 980 nm [145]. Recently, new possibilities have been discovered by nano-engineering the host materials and by developing core-multi-shell formations with independent doping of individual shells. These advances have opened new possibilities in terms of increased light penetration depth, limited local overheating, and multicolor emission capability [155].

However, most fluorescence dyes may not be single-photon excited at these wavelengths (Fig. 4B). They need to be twophoton or three-photon [133] excited. Consequently, to realize deep-tissue molecular imaging, it is necessary to have ultrafast NIR excitation sources and nanomaterials suitable for generating contrast in the NIR range. Commonly used femtosecond/picosecond Ti:sapphire lasers have bandwidths of 10 nm/1 nm at 800 nm and transform-limited pulse-widths of 94 fs/0.9 ps. Their operation wavelength is tunable in the NIR range (700–1000 nm) for the two-photon excitation of most blue, green, and yellow fluorescence or phosphorescence dyes. The advantages of two-photon excitation in the NIR range is reduced out-of-focus photo-degradation of fluorescent dyes during 3D-sectioned imaging. They also serve as pump waves in CARS and SRS microscopy. Their application to different optical contrasts has been widely reported elsewhere. To excite red fluorescence or realize deeper imaging depths, 1000–1300 nm ultrafast light sources would be highly desired. Some researchers use Ti:sapphire lasers to pump an optical parametric oscillator (OPO) to obtain a tunable 1050–1300 nm femtosecond source. Nevertheless, such a setup is usually expensive and the maintenance of this system is complex and time consuming. In this section, we will review new advances in the 1000–1300 nm NIR laser sources for *in vivo* deep tissue imaging.

3.1. Nd:YAG and Nd:YVO₄ lasers

The neodymium-doped yttrium aluminum garnet (Nd:YAG) laser is a four-level laser system operating at ~1064 nm using Nd³⁺ ions as gain centers. The upper state fluorescence lifetime is long (~230 μ s) [156], so that a significant population inversion can be maintained with a relatively low pump power. Therefore, it is usually operated as a Q-switched laser generating nanosecond pulses with a pulse energy on the order of tens of millijoules. Such a high pulse energy has been employed for deep tissue NIR photoacoustic imaging [157]. With the help of OPO, the wavelength of a Nd:YAG laser can be tuned to 1210 nm and 1700 nm for CH₂ and CH₃ bond-selective photoacoustic imaging, respectively [158]. With Nd³⁺ ions doped into yttrium vanadate (YVO₄), the gain bandwidth increases to 1 nm, which allows the generation of 2-ps-long laser pulses [159]. Laser sources of this type can be used for generating Stokes waves in CARS microscopy. The 532 nm SHG output of this laser can synchronously pump another OPO to generate tunable pump waves [160]. For CARS microscopy applications, picosecond lasers should have sufficiently strong nonlinearity for generating anti-Stokes signals. Femtosecond lasers will increase the non-resonant background of four-wave mixing and reduce the CARS contrast.

3.2. Cr:forsterite lasers

Pumped by 1064 nm Nd:YVO₄ or Yb:fiber lasers, the ultrafast Cr:forsterite laser can produce 1250 nm femtosecondduration pulses for minimally invasive SHG, THG, and TPF microscopy [92–95]. Compared with Ti:sapphire lasers, the operation wavelength is far from the two-photon excitation wavelengths of most endogenous pigments such as NADH or flavins [161], which significantly reduces the on-focus damage and unwanted auto-fluorescence. This explains why Cr:forsterite lasers can increase the excitation intensity for generating THG without incurring tissue damage. Owing to these least invasive properties and highest penetration depth of wavelengths, Cr:forsterite lasers have been widely used in preclinical [92–95,105] and clinical studies [104,111,112,116,162–164] *in vivo*. Liu and co-workers employed this laser to develop a harmonic generation microscopy for the virtual optical biopsy of tumor microenvironments and tissue inflammation. The authors tracked collagen remodeling in melanoma microenvironment and extracted quantitative features for diagnostics [105].

3.3. Yb:fiber lasers

With a fiber amplifier, the ytterbium-doped fiber laser can produce 1030 nm pulses with a 300-fs pulse width and 40-nJ pulse energy. Using fiber-based OPO, this light source can be used to perform TPF and CARS multimodal microscopy [165]. If

the operating pulse energy of the Yb:fiber laser is reduced to the 1 nJ energy level (e.g., using dispersion compensation instruments), the pulse width can be optimized to 30 fs and the laser can be used for TPF, SHG, and THG microscopy [166].

3.4. Soliton-self-frequency shifted Er:fiber lasers

Tunable ultrafast lasers such as Ti:sapphire can only operate around a single wavelength. For multiple label imaging, researchers typically tune the wavelength for each label or choose a compromise wavelength that can excite them all. This arrangement induces bleed-through problems and reduces the contrast of fluorescence microscopy. Besides, for fast biological events such as neuronal action potentials or cellular circulation, multiple wavelength NIR femtosecond sources are required. To resolve this problem, we used a 400-nJ-energy, 1550-nm-wavelength Er:fiber laser to excite a large modearea photonic crystal fiber and generate multiple solutions at 1900 nm and 1728 nm through a soliton-self-frequencyshift (SSFS) mechanism. After SHG, we obtained 775 nm, 864 nm, and 950 nm femtosecond sources for multiple labeling TPF imaging [167]. Using this SSFS mechanism, by choosing larger photonic crystal rods, the 1550 nm source can be efficiently red-shifted to generate intense 1700 nm light sources and thus realize deep-brain three photon fluorescence microscopy [133].

3.5. Laser diodes

Fig. 4D shows that many semiconductor-based materials can be used as light sources. Although the tunable range spans 630–900 nm (GaInAlP and GaAlAs), 710–850 nm (AlGaAs/GaAs), 900–1000 nm (InGaAs/GaAs), and 1000–1650 nm (InGaAs// InP), laser diodes, which are commercially available and sufficiently powerful ($P \ge 1$ W) for biomedical theranostics applications, are limited to several discrete wavelengths that fall in the optical windows of the skin [131,132]. Most frequently, laser diodes operate in the 635–670 nm, 780–830 nm, 905–915 nm, 920–980 nm, 1064 nm, 1260 nm, and 1550 nm wavelengths, with an optical power above 1 W. Other wavelengths are not easily available, with a power of ~100 mW.

4. Smart NIR linear and nonlinear optical nanomaterials

4.1. NIR linear optical nanoparticles

Using linear optical materials, fluorescent molecules in the visible range (~400-650 nm), such as fluorescein dyes, rhodamine-related dyes and others [168] have been widely used in bio-sensing, immunoassays, Western blot detection and high-throughput devices., but have been less utilized for in vivo measurements. However, indocyanine green (ICG)encapsulating polymers (e.g., polylactic-co-glycolic acid [169] and poly(allylamine hydrochloride) [170]), -micelle [171,172], -lipid [173], -human serum albumin [174], -mesoporous silica [175,176], -silica/Au [177], silica-poly(ε-caprolac tone) [178] and -gold nanomaterial [179] composites are available for NIR brightened and robust bio-imaging detection and for photo-thermal energy conversion. The stability of ICG dyes can be improved by embedding them into polymers and inorganic nano-capsules. Another example is the IR-820 dye encapsulated in 1,2-Distearoyl-sn-glycero-3phosphoethanolamine (DSPE)-mPEG500 polymer nanoparticles through a hydrophobic-hydrophilic self-assembly interaction method developed by Chu et al. [180]. These copolymer nanoparticles are much brighter than the aggregates of IR-820 powder (aggregation-caused quenching). By using a red excitation light near the NIR-I window, blood vessels at the depth of 500 µm could be visualized to obtain 3D reconstructed images of both the vasculature and brain signaling *in vivo*. A new bis(propylthio)tetrathiafulvenyl[i]dipyrido-[3,2-a:2',3-c]phenazine (TTF-dppz) compound with $\lambda_{abs} = 750$ nm and λ_{em} = 975, 986, 1009, and 1020 nm was developed by Lapadula et al. [181]. After conjugating the Yb(III) molecular complex to the surface of silica nanoparticles (~100 nm), this silica-based fluorophore absorbed and emitted in the NIR region $(\lambda_{abs} = 750 \text{ nm}, \lambda_{em} = 983 \text{ and } 1050 \text{ nm}).$

Notably, fluorescent organic nanoparticles that consist of conjugated polymer dots [182,183], conjugated polyelectrolyte dots [184–186], conjugated carbon nanodots [187], or polymer-encapsulated dye molecules [188–190] have attracted considerable attention for bio-imaging applications, owing to their relatively high fluorescence emission quantum efficiency, photo-stability, and low cytotoxicity. However, only a few reports have successfully demonstrated the emission of organic nanodots that emit fluorescence in the NIR wavelength regions [191], which has limited their practical applications for bio-imaging of deep tissue. The excitation and emission peaks within the far-red/NIR wavelengths are highly desirable and match the tissue-transparency window for targeted *in vivo* fluorescence imaging and cancer diagnostics. Bioluminescence does not require optical excitation [192], but its spatial resolution is poor owing to tissue scattering. Another strategy is to employ the NIR absorption of conductive polymers in organic electronics and organic solar cells [193,194], such as poly-pyrrole [195] or poly-(3,4-ethylenedioxythiophene):poly(4-styrenesulfonate) (PEDOT:PSS). They provide strong photoacoustic contrast, achieve effective photo-thermal ablation [196], and have organic stealth for long-term circulation.

In addition to the aforementioned organic nanoparticles, yielding linear response functions in the NIR region, carbon nanotubes bear the promise for improving the efficacy of photo-thermal cancer therapy and the positional accuracy of treatment under the guidance of optical imaging, owing to the unique electronic states assisting electronic transitions in the UV– visible-NIR regions. The fluorescent contrast of dye label-free carbon nanotubes can be excited at 395 nm [197], 488 nm [198], and 660 nm [199]/808 nm [200], yielding emission bands at 485 nm, 530 nm, and 900–1400 nm, respectively. Therefore, researchers have validated that laser light at 785–1100 nm can damage cancer cells after cellular uptake of singlewalled carbon nanotubes (SWNTs) [201–203]. However, two major drawbacks associated with the use of SWNTs are related to their effects on the human body [204]. One issue is the toxicity of SWNTs with respect to cells and organs, which remains controversial [205,206]. Another limitation is the water dispersion of SWNTs in physiological environments.

4.2. NIR nonlinear optical nanoparticles

In most cases, inorganic solid samples can promote THG owing to a large difference between the refractive indices of particles and media (e.g., air and liquid) [207–209]. Metal oxides are characterized by high refractive indices, compared with both water and physiologically relevant media (n = 1.33). In addition to the intrinsic refractive index difference, the thirdorder nonlinear optical properties of sol-gel-derived V₂O₅, Nb₂O₅, and Ta₂O₅ thin films are primarily dominated by the lengths of metal–oxygen bonds. In fact, Tadanori et al. reported that transition metal oxides with the smallest l_b exhibit the highest third-order nonlinear susceptibility $\chi^{(3)}$. Conversely, non-transition metal oxides yield high $\chi^{(3)}$ as a result of a large l_b [208]. In THG microscopy coupled with nanoparticles, the THG signal under intense illumination converts three photons into one photon with a wavelength equal to one-third of the incident wavelength, and decreases the background owing to the cell auto-fluorescence.

Non-centrosymmetric metal oxides with harmonic generation properties have been extensively investigated [210]. ZnO is an n-type semiconductor that has a band gap of ~3.37 eV. The linear and nonlinear PL behaviors of ZnO nanoparticles are related to the intrinsic direct band gap and oxygen-related defects associated with the surface trapping states [211,212]. Owing to its anisotropic crystal structure, ZnO has been used for frequency conversion in SHG microscopy with an amplified Ti:sapphire laser at 800 nm (80 fs, 2.0 W, and 1 kHz) [207]. The enhancement of SHG signals is affected by the lattice atomic structure [213] and polar orientation [214].

Various multiphoton nonlinear optical processes have been studied in Cd-based QDs [119,215]. However, relatively little attention has been paid to the generated multiphoton fluorescence signals of graphitic carbon nanodots [216]. Eu-doped TiO₂ hollow nanoshells provide a new concept for two-photon fluorescence microscopy imaging of HeLa cervical cancer cells using a Ti:sapphire laser at 705 nm (3 W and 120 fs) [217]. The energy relaxation from the two-photon excitation of TiO₂ to Eu³⁺ ions contributes to the red emission at 617 nm. In contrast, depositing Eu ions in the shells of KTiOPO₄ single-crystal nanoparticles yields a dual light-emission property. The emission bands of as-obtained core-shell nanoparticles can be easily tuned to generate SHG from the KTiOPO₄ core (femtosecond laser λ_{ex} : 990 nm, pulse duration of 100 fs, repetition rate of 86 MHz, average power of 1 mW) and red photo luminescence (PL) from the shell [continuous wave (CW) laser λ_{ex} :532 nm, 10 mW] [218].

Two-photon fluorescent probes comprised of a two-photon fluorophore 4-(bis(4-(4-(diphenylamino)styryl)-phenyl)ami no) benzaldehyde [219], phenyl thiourea linker, and amino triphenylamine dendron chelated, exhibited efficient TPF detection of Hg²⁺ in a wide dynamic range of concentrations (5 nM to 1.0 μ M) [220]. Organometallic compounds (e.g., cyclometalated platinum (II) complexes) can be used for two-photon emission live-cell imaging [221,222]. Besides, Pt-based molecules have been also established as anticancer drugs that work by intercalating DNA [223–225].

Regarding organic dyes, molecules such as 1,1,2,3,4,5-hexaphenylsilole (HPS) and bis(4-(N-(1-naphthyl)phenylamino)-phenyl)fumaronitrile (NPAFN) can be loaded into polymeric micelles to form nanocarriers. These nanoconjugates exhibit good protection of the hydrophobic dye and provide high fluorescence intensity for imaging live cells with a low toxic impact [226]. The aggregation of aromatic dyes via the π - π interaction aids in the generation of a strong fluorescent intensity.

Ultra-bright organic dots, consisting of 4,7-bis[4-(1,2,2-triphenylvinyl)phenyl]benzo-2,1,3-thiadiazole (BTPEBT) aggregates, exhibit an aggregation-induced evolution of TPF [227] (Fig. 5). By using a femtosecond Ti:sapphire laser with λ_{ex} = 800 nm, the aggregation-induced emission of BTPEBT can be applied to observe smaller capillaries in 3D imaging of the brain, bone marrow, and ear skin.

Similar aggregation-induced two-photon emission was also observed in nanoparticles using 9,10-bis[4'-(4''-aminostyryl) styryl]anthracene (BDSA) derivative [228,229], pyran derivative, distyrylanthracene derivative [230], perylene-3,4:9,10-tetra carboxylic bisimide [231], and 2-(2,6-bis[(E)-4-(diphenylamino)styryl]-4H-pyran-4-ylidene)malononitrile [232]. These organic dyes in silica composite produced nano-sized hybrids and allowed for cancer cell imaging combined with the indirect excitation of a photosensitizer through two-photon excited energy transfer. Coating this NIR organic moiety on nanomaterials can improve the performance of multiphoton nonlinear optical processes. For example, the surface of Au nanorods coated with NIR polypyrrole (PPY) allows the flux of hot electrons to PPY to perform intracellular TPF imaging. This nonlinear process can be combined with two-photon excited photo-thermal therapy for treating HeLa cells using an 880-nm-wavelength laser with the fluence of 0.86 J/cm² [233].

4.3. Multiphoton up-conversion of lanthanide nanoparticles

Another strong advantage of lanthanide nanoparticles is the nearly perfect photo-stability of lanthanide-doped bio-labels. Neither photo-blinking nor photo-bleaching has been observed in these materials, which implies these can be used for long-term observations, such as studying time-dependent processes or studying the ability to trace such bio-functionalized labels circulating within living organisms.



Fig. 5. (A) Schematic of aggregation-induced emission (AIE) dot fabrication. (B) TEM image of AIE dots. (C) UV-visible absorption and PL spectra of AIE dot suspension in water; $\lambda_{ex} = 425$ nm. (D) Two-photon absorption spectra of AIE dots and QD655 in water and Evans Blue in saline with 0.175 mg/mL bovine serum albumin (BSA). Data are presented as mean ± standard deviation (SD), (*n* = 3). (E and F) Wide-field (40 µm × 40 µm) luminescence images of single AIE dots (E) and QD655 (F). (G) Representative luminescence intensity time-traces for AIE dots and QD655. $\lambda_{ex} = 488$ nm for (E, F, and G). (B) Intra-vital TPF imaging of AIE-dot-stained blood vessels in different organs. (A–C) A time-lapse image sequence of maximum intensity projection showing blood vessels in the brain taken at 0 (A), 15 (B), and 30 (C) min post-injection of the AIE dots. (D–I) Images at various vertical depths of the bora marrow and ear skin. Blue: second harmonic generation; collagen in dermis. Scale bar: 50 µm. $\lambda_{ex} = 800$ nm. Signal collected at 542 ± 27 nm. Reproduced with permission [227].

However, numerous additional features of these materials suggest interesting possibilities. First of all, lanthanide ions exhibit narrow- and multi-band absorption and emission. As a consequence, they exhibit large Stokes shifts, which helps to separate lanthanide emission from a much stronger excitation laser line. In addition, the narrowband and multiband emission makes multiplexing feasible, because numerous spectral codes may be designed for labeling multiple biological targets in the same samples, for example for marking a range of organelles in a single cell to unravel complex morphology, for simultaneously studying a few biological processes with, for example, a luminescence resonant energy transfer (LRET) technique, or for enhancing high-throughput screening performance to detect multiple rare cells or disease markers in human samples. The emission lines from Er, Tm, Ho, Sm, Tb, and Eu overlap partially, but spectral decomposition allows one to distinguish between the different spectral codes (Fig. 6). Although the locations of absorption/emission bands do not vary in terms of the wavelength, as can be found in quantum dots or organic dyes (only a relatively small variation in the spectral location and subtle structure of bands is observed across different host matrices), the color variation of Ln³⁺-doped labels has been realized by engineering energy transfer pathways either passively (by varying the size [234,235], shape, composition, morphology [236,237], host matrix [238,239], surface ligands [240]; by admixing optically inactive ions such as K⁺, Li⁺ [241], Fe³⁺ [242]; or by substituting Gd^{3+} for Y^{3+} in NaYF₄ [243]) or actively (by varying the relative concentration of Ln^{3+} dopants [244-246], adding spectrally active ions such as Mn²⁺ [247,248] or Ce³⁺ [249,250]). Hierarchical layer-by-layer nanostructuring has been also demonstrated to yield a simple cascade multicolor emission [251]. Alternatively, more flexibility in designing optical codes can be achieved either by homogenous mixing of lanthanide complexes inside bar-codes (by mixing defined different color UCNPs within single SiO_2 or polyethylene glycol (PEG) beads [252]) or by doping lanthanides independently into cores and shells in active-core-active-shell nanoparticles [88]). Interestingly, owing to the relatively long luminescence lifetimes of lanthanides, the ability to intentionally design optical codes in the time domain has been previously predicted [253] and demonstrated [254,255].

The features of lanthanides become even more interesting when it comes to biosensing, and owing to multiple upconversion emission spectra spanning the visible and NIR spectral regions, ratiometric biodetection becomes possible at increased depths; one of the emission bands serves as a reference, while the other is modulated proportionally to the concentration of the analytes. For example, LRET biosensors have been demonstrated, such as DNA hybridization or enzyme activity biosensors. In addition, pH- or [Hg]-sensitive probes were designed based on such ratiometric detection. This technique relies on the donor's emission quenching (DQ), non-radiative LRET, or inner filter effect (IFE), where only one of many Ln³⁺ up-converted emission bands (e.g., the one at 470 nm of Tm³⁺ emission) overlaps with the absorption of an acceptor (DQ, LRET) or sensitive chromophores (IFE), while the other emission bands (at 650 nm or 800 nm of Tm³⁺ emission) remain unaffected. Since the chromophores used in such bio-sensors are capable of enabling biological recognition (DQ, LRET) or responding to changes in the local chemical environment (e.g., changes in ionic concentrations) by changing the absorption



Fig. 6. Multicolor emission of (a) NaGdF4:Yb,0.5%Tm, x Er@NaGdF4:Yb@NaNdF4:Yb, (b) NaGdF4:Yb,0.5%Tm, xEr@NaGdF4:Yb,15% Eu@NaNdF4:Yb and (c) NaGdF4:Yb,0.5%Tm@NaGdF4:Yb, xEu@NaNdF4:Yb colloidal solution of up-converting active-core-active-shell lanthanide doped nanoparticles [88]. (d and e) show illustrations of up-conversion and the corresponding Commission Internationale de l'Eclairage (CIE) coordinates, obtained for colloidal solutions of active-core-active-shell UCNPs. Reproduced with permission [88].

spectra (IFE), only one emission band is modulated, while the other serves as a reference for quantitative measurements. Examples of such ratiometric biosensors include pH [256], carbon dioxide [257], ammonia [258], mercury [259], glucose [260], cyanide anions [261], hydroxyl radicals [262], and oxygen [263] sensing. A similar idea was employed by Kang et al. to study the release characteristics of the ibuprofen drug. The up-conversion emission quantum yield was proportional to the amount of released ibuprofen [264], generating a platform for drug delivery and drug release monitoring.

The most severe drawback of Ln nanoparticles is a relatively low quantum efficiency owing to their low absorption and emission cross-sections, which result from a forbidden f-f optical transition. The efficiency, although much lower than that of organic dyes or quantum dots, is not prohibitive for bio-applications, and ultrasensitive immunoassays and detection of a few cancer cells have been successfully demonstrated. This low efficiency is also the outcome of a large surface to volume ratio of nanomaterials. A substantial number of doping ions are located close to the NC surface and their excited states are thus susceptible to chemical microenvironment, nanocrystal defects, ligands and solvents, which have been recognized as quenching mechanisms. However, these side effects may be relatively easily reduced using core-shell lanthanide-doped nanoparticle architectures. It has been shown, that ~4-nm-thick passive (un-doped) shells are suitable for complete surface passivation and for reducing the susceptibility to the nanoparticles' surface chemistry or to solvents [265]. A number of methods have been sought for improving the up-conversion emission intensity in nanoparticles. Basically, (i) surface passivation [266], (ii) passive (K⁺, Li⁺)/active ion (Nd³⁺, Mn²⁺) co-doping [247,248,267,268], (iii) plasmonics [269], (iv) increasing the concentration of the sensitizer (Yb³⁺) [270], (v) host selection [271], and recent smart active-core-active-shell nanoparticle designs [272–274] are used to achieve this goal, with different outcomes. The photo-physical properties of lanthanide-doped nanomaterials have been extensively reviewed [275–278]. In addition, the biomedical properties and applications have been discussed in a growing number of excellent review articles [90,275,279–288].

Owing to the numerous photo-physical advantages of UCNPs that were briefly reviewed above, lanthanide-doped nanoparticles have been used in (i) passive, (ii) modulating, and (iii) active biomedical applications (Fig. 7) [288]. Passive applications include the use of the nanoparticles as contrast agents in fluorescence microscopy, MRI (owing to the accumulation of Gd³⁺ ions, e.g., NaGdF4 matrix [289–292]), X-ray imaging, or PET imaging (owing to the F¹⁸ isotopes attached to the nanoparticles' surface [293,294]). Often, these imaging modalities can be combined in pairs [292,295–298] or triplets [299]. Recently, hexamodal imaging has been demonstrated with PoP-coated UCNPs, which combined CT, PET, up-conversion, Cherenkov luminescence, photoacoustics, and FL [20]. The fluorescence contrasts of UCNP labels were so sensitive that as few as 10 stem cells could be detected *in vivo* [300] for at least one week after delivery [301].

Active applications include methods in which UCNPs affect biological tissue; examples include the possibility of hyperthermia (local overheating of cancerous tissue) with UCNPS directly [302,303] or with UCNPs bound to Au/Fe₂O₃ nanoparticles [304–306], (chemo-, geno-) therapeutic drug delivery [307], or up-converted photodynamic therapy [308–313]. UCNPs



Fig. 7. State-of-the-art biomedical applications of lanthanide-doped (up-converting) nanoparticles. Active, passive, and modulation applications of UCNPs, which respectively relate to the direct impact of UCNPs on surrounding tissue (through temperature or PDT activation), lantern type contrast agents (PET, MRI, fluorescence), and modulation of spectral properties of UCNPs by either environment (e.g., CO₂, pH; (a)) responsive bio-molecules (hybrid sensors), FRET-based (quenching (b), nucleic acids hybridization (c), enzymatic (d)) sensors, or analytes (direct analyte sensors, e.g., temperature (g), drug (e), H₂O₂ (f)) [288].

are not very efficient in causing hyperthermia owing to a relatively weak absorption of light by lanthanides and thus, a relatively low heating efficiency. UCNPs are also not very well suited as drug delivery agents, because a crystal (and thus solid and firm) matrix is necessary to host lanthanides for making them luminescent. For this reason, there is no way to introduce external compounds into a system containing UCNPs. UCNPs can, however, be covered with mesoporous SiO₂ or polymeric shells, which may carry and release drugs. Prolonged and controlled release of drugs may be of high importance. LaF₃:Yb³⁺, Er³⁺/nSiO₂/mSiO₂ microspheres [314] and NaYF₄:Yb³⁺,Er³⁺/nSiO₂/mSiO₂ [264] microspheres were demonstrated to sustain and control the release of ibuprofen that was loaded into mesoporous shells of UCNPs.

The most spectacular and promising example of using UCNPs in theranostics is NIR-initiated photodynamic therapy, also termed up-conversion photodynamic therapy. Such hybrid up-conversion-PDT nanoparticles include UCNPs as luminescent bio-probes in the core, and simultaneously photosensitizing molecules (e.g., chlorin e6) are either covalently attached to the surface of nanoparticles or embedded in mesoporous SiO₂/PEG shells. In most of these applications, UCNPs act as light transducers, converting light from NIR to visible or NIR spectral region, which basically increases the penetration depth of the excitation light and thus enables photo-biosensing [315,316] and phototherapy [305,317–319] of heterogeneous tissues. Moreover, UCNPs are well suited for theranostic applications [320,321], because it is relatively easy to combine several functional features within individual nanoparticles.

5. Application fields of NIR nanomaterials in cancer theranostics

Conventional cancer treatment modalities, such as surgery, radiation therapy, and chemotherapy, all result in serious side effects and in many cases (especially for rapidly developing cancers or tumors within delicate tissues in the head and neck), do not achieve complete removal of the cancer cells. For this reason, a single underlying biological process that could allow for selective targeting and destruction of diseased cells while preserving their healthy functional neighbors is highly desirable. However, the problem of successful cancer treatment originates from the complex nature of cancer development. Cancer cells originate from the host, exhibit un-regulated proliferation, and are often found to migrate [322]. Moreover, individual tumors display heterogeneous properties in terms of structure, biochemical behavior, nature of the surrounding microenvironment, and susceptibility to their biochemical environment and susceptibility to treatments. Despite the conceptually interesting idea of using antibodies to target drugs to cancerous tissues and cells [323–326], very few significant successful attempts have been made to date. This is largely for the following reasons [327]:

- (1) Difficulties in achieving tumor-specific antibodies that also display high affinity.
- (2) High biochemical (i.e. antigen) heterogeneity of tumors throughout their mass.
- (3) At the organ scale, the antibodies either do not reach the tumor site or do not easily penetrate the tumor, and therefore remain in the tumor vasculature.
- (4) At the cellular scale, the antibodies are generally not designed to penetrate the tumor cells and thus the cytotoxic agents may not reach the most sensitive intracellular sites (such as the mitochondria).

To investigate and understand these problems in the microenvironment of a specific biomedical context in an *in vivo* setting, one needs molecular probes with NIR optical contrasts. One also needs to design an appropriate imaging method for NIR excitation and detection. Once this is achieved, cancer cell distribution, the microscopic pharmacokinetics of nanomedicines, delivery of therapeutics, cell responses, and cell-cell interactions can be visualized and analyzed. This integration of NIR nanomaterials (i.e., ICG-encapsulating polymer [169–174], conjugated polymer dots [182–186], luminescent carbon nanomaterials [201–206], biocompatible Ag₂S nanoparticles [42,328], aggregated induced emission (AIE) dots and nano-oxides [211–214,216,217], dyes [219–227], plasmonic nanoparticles [329–338], upconverting lanthanide nanoparticles [234– 240,242–255], and upconverting lanthanide nanocomposites [289–314]) with the NIR imaging system provides a proofof-concept platform and visual evidence critical for the success of translational cancer nanomedicines.

5.1. Biosensing assays

Understanding the physical and chemical features of the tumor microenvironment provides an insight for the design of nanomedicines with better targeting and delivery efficiencies. The physiological parameters of interests include the partial oxygen pressure pO_2 , pH value, and glucose level, permeability of the blood vessels and ROS levels. To visualize the conditions and dynamic processes *in vivo*, conjugated polymer and silicate nanoparticles have been developed as important molecular probes with NIR optical contrasts. For example, phosphorescence probes that can detect pO_2 [31–35] are useful in understanding the correlation between a hypoxic environment and the angiogenesis process [339,340]. In the material chemistry of organic light-emitting diodes, transition-metal based phosphors and molecular complexes have been developed as efficient phosphorescence chemicals [341,342]. Their strong spin-orbital coupling allows for efficient phosphorescence, which allows them to serve as sensors of pO_2 [343]. Among them, metalloporphyrin complexes [339,340,344], Ruthenium (II) complexes [345] and Iridium(III) complexes [346] have been designed for pO_2 sensing and imaging. For deep tissue pO_2 measurement, additional dyes like coumarin-343 have also been incorporated as two-photon antenna to raise the phosphorescence quantum yield [347].

Surface passivation is critical for these phosphorescence probes in order to avoid self-aggregation or adsorption to biomacromolecules [33], although this may alter their oxygen sensitivity and result in erroneous measurements. Regarding the hypoxic condition in the tumor microenvironment, it is well known that this physiological property can change the glucose metabolism of cells from oxidative phosphorylation towards lactic acid fermentation, a less efficient way for cell to obtain energy currency, namely adenosine triphosphate (ATP). In this case, cells will take up more glucose to produce the same amount of ATPs and therefore become more acidic. This shift in acidity can be quantified using pH-sensitive fluorescent dyes [348]. Carried by pH-low insertion peptides, pH-sensitive fluorescent peptide probes can specifically measure the pH value at the cell surface [349]. Using dextran loading, Cong et al. have also developed a pH-activated NIR fluorescent probe (polylysine-liked rhodamine/IR783/PEG/In³⁺-DOTA/dextran). These designs show that peptides or dextran could serve as targeting carriers for the sensing of ions around tumor cells [350].

The hypoxic tumor can further induce angiogenesis, producing a vasculature with an abnormal network. The endothelial surface of the vasculature is often fenestrated with gaps between endothelial cells, due to a decrease in the number or adhesiveness of pericytes [351]. These gaps can however enhance the permeation and accumulation of nanomedicines in circulation. To evaluate the particle size that results in maximal accumulation, researchers have commonly used fluorescent dextrans with molecular weights ranging from 20 to 2000 kDa [352,353]. To achieve deep tissue analysis of vessel permeation, NIR macromolecular dye-tagged dextrans therefore hold great promise.

Many proof-of-concept biosensors have been developed using UCNPs combined with bio-responsive molecules. While the UCNPs are not susceptible to changes in the chemical or bio-environment, they may be excited at much greater depths within the tissue mass. Due to multi-band emission and the overlap of lanthanide ion emission with the environment-sensitive molecules anchored at the surface of the UCNPs, ratio-metric bio-sensing using IFE can be achieved and sensitivity can be enhanced. Such types of biosensors have been shown to be effective in studying immunochromatographic assay (e.g. *E. coli* [354] and human chorionic gonadotropin [355] detection), DNA-hybridization [356,357], enzyme activity [358], pH value [256], carbon dioxide [257], ammonia [258], mercury [259], glucose [260], cyanide anions [261] or oxygen [263] concentrations. These sensors expand the well-known properties of some organic dyes with the ability to read the biosensor response at NIR photoexcitation wavelengths.

In the context of photodynamic therapy, it is of utmost importance to understand whether ROS are actually released around tumors. The use of organically modified silicate (ORMOSIL) to load ROS sensitive dyes and a reference dye to perform ratiometric fluorescence measurements is common [359–362]. The polymer matrices used can inhibit interaction of the dye with intracellular proteins, protect the dye from degradation, and inhibit undesired sequestration into subcellular compartments. A common strategy is to design systems where the fluorescence quenching mechanism occurs on the surface of metallic nanoparticles [363,364]. As long as the ROS interact on the surface-coating of ligands on metallic nanoparticles, the fluorescence will be quenched. Therefore, these types of nanoprobe can be used to sense ROS level in cells with excellent spatio-temporal resolution.

5.2. NIR imaging methods

Optical imaging relies on the illumination of light waves on a subject, the generation of contrast by molecules within, and the mapping of responses by an optical system and cameras. Therefore, the best-achievable resolution and imaging depth of an optical imaging system will be determined by the illumination and collection methods. Depending on the desired depth and the scattering properties of tissues, the types of optical imaging system can be divided into either diffusive or ballistic. For large-scale and whole-body imaging, nanomedicines usually reside in tissues at a depth several times the scattering lengths of light. The signal photons can be scattered multiple times, causing a loss of coherence of the wave-front before they leave the turbid tissue environment. In this case, the typical resolution of diffused photon imaging will be on the order of a centimeter. For in vivo microscopy of complicated tumor microenvironments, a ballistic imaging system with sectioning capability should be adopted. In this case, nanomedicines are illuminated by a focused laser beam and the imaging depth will be well-within the scattering length of light. For the NIR light source, the ballistic length is 300–500 µm for skin [131] and 1.5 mm for embryos [365]. The sectioning capability can be achieved using either a confocal imaging method, laser-scanned nonlinear optical microscopy, or light-sheet illumination. The resolution of a ballistic imaging system is typically sub-micron, so it can reveal the structural and sub-cellular details of the tumor microenvironment. For imaging of tumor vasculatures, the imaging depth should be deeper than the scattering lengths. To obtain a sufficiently high spatial resolution for vasculature imaging, the photoacoustic contrast should be considered. Since acoustic waves have less attenuation and less diffraction than optical light, the lateral resolution of photoacoustic imaging is typically 40 µm at 3-mm imaging depth. This resolution is much better than that of diffused optical imaging and can easily map tumor vasculatures.

To achieve low energy loss (i.e. prevent absorption and scattering at short wavelengths) and deposition in biological tissues, anisotropic Au particle [366], Ag₂S dots [42,328], CNT, dye-loaded composites [169–174], and nonlinear nanomaterials [211–214,216,217], upconverting lanthanide nanoparticles [234–240,242–255,289–314] have proven to be satisfactory choices of contrast agents for providing emission in the long wavelength region after long wavelength excitation in deep tissues.

5.3. NIR fluorescence for image-guided surgery

Various radiological imaging modalities (CT/MRI/PET) are used as a preoperative assessment for surgery guidance. During surgery, NIR contrast agents may be administered intravenously or intraperitoneally and visualized using an NIR fluorescence imaging system with adequate NIR excitation light, collection optics, filters, and a camera sensitive to NIR fluorescence emission light [367] (Fig. 8). For higher resolution diagnostics or intra-operative navigation, a microscopy imaging modality is required. For example, the biopsy of sentinel lymph nodes (SLN) around tumors is a critical diagnostic procedure used for the typing and staging of tumors. The administration of NIR fluorescence agents *in situ* combined with large-scale fluorescence imaging can help doctors visualize SLN [368]. In clinical practice, the *Food and Drug Administration*-approved ICG has been used off-label for the tracing of SLN. ICG excitation and fluorescence wavelengths are 780 and 822 nm, respectively, which is beneficial for deep-tissue real-time imaging. To increase solubility and fluorescence yield, a common strategy relies on the adsorption of ICG with human serum albumin. However, compared with ICG, type II semiconductor QDs can provide brighter 850 nm fluorescence for SLN imaging *in vivo* [369]. To provide favorable accumulation and longer retention of the NIR contrast agent in the draining SLN, the appropriate particle must be selected (20–50 nm) [370]. For deeper SLN imaging, several photoacoustic contrast agents have been developed. Perfluorocarbon nanoparticles loaded with NIR fluorescence dyes can achieve simultaneous NIR optical and photoacoustic imaging of SLN *in vivo* [371]. Whole-body lymph nodes have been visualized using a combination of NIR excited photoacoustic imaging and semiconducting polymer nanoparticles [372].

In the surgical resection of tumors, clinicians commonly used larger boundaries around the major nodule to maximize the clearance of cancer cells. But for organs like the brain, extra resections may result in a loss of psychological and motor functions. Besides this, for cancers with irregular or dendritic shapes, cancerous tissues may be present at the boundary of the excised lesions. Therefore, the residual cancer cells may result in recurrence and affect the prognosis after surgery. Thus, to identify tumor cells from adjacent non-tumor cells and critical structures like neurons intra-operatively, a common strategy is the development of contrast enhancement methods or fluorescence probes to augment the visual differences between normal and cancer cells.

Compared to 10–100-nm sized crystals, a contrast agent designed on a molecular scale (<10 nm) has the benefit of better penetration and uniform distribution at specific sites of the body. Therefore, several extrinsic fluorescence contrast enhancers have been developed for guided surgery [373]. These fluorescence probes can either target the metabolic features, or the hallmarks of specific cancer types. For example, the use of 5-aminolevulinic acid results in the accumulation of protoporphyrin IX in glioblastomas [374]. This porphyrin can be effectively excited at the Soret band (an intense peak in the blue wavelength region of the visible spectrum) and imaged at 600–750 nm NIR wavelengths. Another efficient approach is to use protease-sensitive probes, where the red fluorescence of cell-penetrating peptides can report the presence of tumorassociated matrix metalloproteinases (MMPs) [375] and thus the presence of residual tumors. To avoid damage to peripheral nerves during surgery, fluorescent peptides specifically bound to neural cells have been developed for systemic administration [376].



Fig. 8. NIR fluorescence for image-guided surgery. When performing surgery, an NIR contrast agent may be injected and monitored using an NIR fluorescence imaging system, with appropriate NIR excitation laser light and a camera sensitive to NIR fluorescence emission light in order to capture the signals and project them on a standard computer monitor or wall projector. Usually, targets up to 5–8 mm deep can be detected using NIR fluorescence imaging whereas a target deeper than 25 mm would not be detected.

5.4. NIR photo-triggered drug release

For better absorption or therapeutic efficacy, the time and place of drug release needs to be precisely controlled. For example, some pills or capsules carrying drugs have pH-sensitive coatings to avoid digestion in the stomach or intestine, in order to achieve specific colonic drug delivery [377]. To actively control the release of drugs from carriers, excitation using optical [4,455–341], acoustic [378,379] or magnetic [380] energy is required. These methods allow multiple dosages of drug to be achieved from a single administration with precise control of the timing, duration and magnitude. For light-activated drug release, the depth of action can be increased by illumination with NIR light [381]. For example, by exploiting the NIR absorption of ICG [382], the membrane of doxorubicin-loaded red blood cells can be thermally destroyed in order to achieve efficient and specific drug release. The SPR absorption of gold nanostructures can also enhance the local electric field and achieve NIR-triggered drug release [4]. For non-thermal release, a photo-labile linker that covalently bound a drug to dendrimers or dendrons has been developed [383]. To cleave a linker using NIR light, upconverting nanomaterials or two-photon excitation processes [384] can also be used. Changing the physical properties of loading matrices can also achieve controlled release. Successful processes include photo isomerization [385], photo-induced gel swelling [386], photo-reactive molecular valve [387], or the photo-decompression of particle sizes [388].

5.5. Photo-thermal therapy

Dark colored materials can absorb light and convert photo-energy into heat. This approach can be adopted for nanoparticles as means to treat tumors. As the nanoparticles absorb light energy, excited carriers will release their energy either through the emission of photons or through the generation of photons and heat. If the nanomaterial (i.e., Au [366,389], carbon [390,391], oxides [337,338,392–394], or dyes [174,180,382]) lack an efficient photon emission route, by using CW lasers as light sources, then most of the absorbed light energy will be converted into heat. In photo-thermal therapy (PTT), when the temperature of particles rises above 40 °C, adjacent cells will be abruptly damaged through pore formation, or will undergo apoptosis due to heat-shock [395].

Thermal therapies have been used since the 1980s for enhancing human metabolism and treating diseases (e.g. tumors in cancer therapy) [396,397]. In fact, the different thermal gradient readily changes tissue elasticity and blood flow rate [398,399], as well as inducing cell death pathways [400,401]. However, the thermal effect on cells in a changing microenvironment remains unclear because of the systemic host effects [402]. To avoid this uncertainty, controlled and localized heating is required for cancer treatment. For example, a carbon nanotube has been reported as an NIR-II light-to-heat converter for PTT of malignant cells [390,391]. The design of a multi-branched Au structure (~350 nm) resulted in a broadened absorption band extending to NIR wavelength [389]. Using a 1064 nm CW laser, both photo-thermal and photo-dynamic therapies have been achieved at a very low power intensity of 130 mW/cm². By special design, luminescent, lanthanide-doped, nanoparticles can serve as hyperthermia agents. For example, heavily doped Nd³⁺ can be used as nano-heater [303], imaging agent, and nano-thermometer [403], for remote tracing of temperature during hyperthermia. In addition to common plasmonic metal nanoparticles, a recent review manuscript has revisited the classification of NIR-absorbing non-metal nanomaterials for photo-thermal applications *in vitro* and *in vivo* [404].

5.6. Upconversion induced photodynamic therapy

Photodynamic therapy (PDT) was introduced 100 years ago [405]. This technique was used to treat various cancers, agerelated macular degeneration, and actinic keratosis [406]. The activation of PDT requires spatial co-localization of three elements: light, a photosensitizer (PS), and oxygen. For a photosensitizer molecule such as porphyrin, part of the excited electrons may couple to triplet states through intersystem crossing. These long-lived triplet electrons may further produce ROS like singlet oxygen or free radicals. These ROS can destroy tumor cells directly, damage the tumor-associated vasculature, or activate an immune response against tumor cells [405]. Individual elements of the PDT procedure themselves are harmless and non-destructive, but as soon as they co localize, the singlet oxygen produced during PDT can oxidize critical cellular macromolecules such as lipids, nucleic acids, and amino acids, thereby inducing alterations in cellular permeability, damage to the plasma membrane, mitochondria and lysosomes [407,408] which in turn lead to cell death by necrosis or apoptosis [409].

Owing to the PS's preferential accumulation in cancerous tissues and cells, these cells are killed with higher spatial selectivity than chemotherapy or radiotherapy, and thus the secondary effects to patients may be considered as negligible. Currently, PDT treatment of numerous cancerous diseases (e.g. early lung cancer [410,411], Barrett's esophagus [412–414], bladder cancer [415,416], head and neck cancers [417], and skin cancers [418–420]) and non-cancerous diseases (e.g. age related macular degeneration [421], bacteria eradication [422]) have been approved.

Despite numerous successes in PDT treatment and numerous approvals for medical use, there are still many issues to be solved. Most of the PSs used to date display only a slight preference for malignant cells, often leading to significant skin photosensitivity and high uptake by healthy cells and tissues. The low uptake contrast between abnormal and normal tissues has stimulated biochemists to design third-generation PSs that are actively targeted towards diseased tissue. Unfortunately, PSs used for PDT have a tendency to aggregate owing to their planar aromatic ring systems, which also allows non-specific binding to bio-molecules (e.g. serum proteins such as albumins, lipoproteins, and high-density lipoprotein) [327]. This may lead

to difficulties in quantifying the biological activity and cytotoxicity of such conjugated and un-conjugated PS molecules, because the photo-physical and photo-chemical properties of such PS variants may differ significantly. In addition, such PS bio-conjugation, or PS-PS interaction, may decrease the PS's absorption coefficient, singlet state lifetimes, and triplet state yields, as well as the PS's excited state lifetimes. This in turn may affect the production of ROS during illumination and thus decrease the photo-cytotoxicity of the dye conjugates. Furthermore, as has been mentioned above, it is necessary not only to consider delivery of the PS to the target cell but also to get efficient accumulation of the PS at susceptible sub cellular locations.

Another severe drawback of conventional PDT treatment that hinders its broad adoption for solid high-volume tumors, and limits its use to only superficial carcinomas in epithelial tissues, is the low penetration depth of light suitable to photoexcite conventional photo-sensitizers. The approved photosensitizers usually absorb below 700 nm, but short-wavelength light undergoes significant scattering and absorption by tissue chromophores. Typically, porphyrin-based photosensitizers [Protoporphyrin (PpIX), Fotofrin, etc.] have a Soret band around 400 nm and a series of Q-bands at the green/red spectral regions [423,424]. Unfortunately, as described in previous sections, light in the 300–650 nm range will be strongly absorbed and scattered by pigments and heterogeneous structures in tissues. The effective region of conventional PDT is therefore rather limited. One approach to solving this issue has been through the use of light diffusers that can be inserted into the tissue to increase the effective volume of PDT treatment. The obvious disadvantage of such an approach is its invasive nature.

In order to overcome these drawbacks, an interesting idea has been proposed to combine photodynamic therapy with upconverting nanoparticles. Under NIR light, these up-converting nanoparticles demonstrate a relatively efficient upconversion to the visible range and may trigger conventional photo-sensitizers (Fig. 9). Such an approach demonstrates some significant advantages over conventional approaches:



Fig. 9. Comparison of conventional photodynamic therapy (PDT) (left) and up-conversion PDT (right hand side) in terms of penetration depth and mechanism. Owing to the short wavelength of light typically required for photosensitizers, the penetration depth and PDT efficiency is usually limited to shallow skin layers. This is as a result of light scattering and absorption by skin/blood components (e.g. collagen, elastin, hemoglobin, etc.). Opposite to PDT, UC-PDT exploits NIR photoexcitation, which penetrates deeper into the skin layers and enables cancer treatment of larger masses. Conventional PDT, occurs by absorption, singlet \rightarrow triplet inter system crossing, followed by free-radical (H₂O₂, OH⁻ or ¹O₂) production, whereas UC-PDT occurs by the up-conversion process and indirect energy transfer to PS. While UC-PDT is less efficient in quantum terms than PDT, the use of NIR enables deeper light penetration and ultimately a more effective treatment.

- 1. Photosensitizer compounds encapsulated in mesoporous silica are protected from degradation in the complex biological environment. In addition, self-aggregation and conjugation to other bio-molecules (such as albumin) does not occur and consequently the PS photochemical properties are preserved. In addition, encapsulation limits their photo-bleaching and biochemical inactivation, thereby maintaining efficient ROS production during illumination leading to efficient photo-induced cytotoxicity.
- 2. Most of the current photo-sensitizers are hydrophobic, thus either polymer [425,426] or silica-based [309,427] shells will enable solubilization of these PSs [as UCNP@shell(PS)] in aqueous buffers.
- 3. The hybrid UCNPs@Shell(PS) NPs, upon bio-functionalization may offer the ability to target the PS to the desired cells. Chatterjee et al. were the first to demonstrate folate receptor targeting using UCNP-PS nanoparticles [428].
- 4. NIR radiation is used to initiate PDT, which offers an increased light penetration depth and suitability to treat larger tissue volumes. A new up-converting nanoparticle design has recently been demonstrated showing novel advancements using an active-core-active-shell design. By using Nd³⁺ primary up-conversion sensitizers, instead of the typical Yb³⁺ sensitizers, excitation light penetration depth was significantly improved. This is possibly a result of the significant reduction (around 25 times) in the absorption coefficient of water at 800 nm for Nd³⁺ vs 980 nm for Yb³⁺. The overheating problem was also simultaneously diminished compared to conventional Yb³⁺-RE³⁺ up-conversion pairs [145,429].
- 5. These advanced hybrid NPs may be designed to release their cargo (e.g. doxorubicin chemotherapeutic agents [307]) upon encountering the decreased pH levels often found in cancer tissues.
- 6. Numerous imaging modalities are also offered by UCNPs@shell(PS), such as up-converted multicolor (and thus multiplexed) luminescence in the visible and NIR spectral region, or MRI imaging (through the use of Gd³⁺ ions within the UCNPs). Additionally, the UCNPs do not photo-blink, and are not susceptible to photo-bleaching, thus allowing for prolonged visualization.
- 7. The shell may be intentionally designed to ensure a sufficiently long circulation time in the blood stream in order to target tissues with a desired antigen profile via a receptor-mediated delivery systems or via an EPR mechanism [317,318].

The majority of the studies reported so far demonstrate a proof-of-concept for UC-PDT experiments. However, further research is required to (i) improve the efficiencies of every single step in the NIR light \rightarrow visible light \rightarrow PS \rightarrow ROS path [430], (ii) design multimodal (PDT, drug delivery, fluorescence/luminescence/PET/MRI imaging) nanoparticles (iii) design smart NPs (e.g. that can release cargo upon an Ab-Ag interaction or upon pH lowering), (iv) bio-functionalize the NPs to achieve long circulation times and allow for highly selective targeting to the tumor sites. Due to the very low quantum efficiency of up-conversion (typically less than 1%) and strong power dependence, further research must be devoted to optimize the photo-physical properties of nanoparticles. One of the most important issues, is the development of methods for characterization that are capable of quantitatively comparing different approaches and that would then allow for optimization of theranostic agents in absolute terms.

5.7. Photo-dynamic therapy and photo-thermal ablation combined with NIR detection

The remarkably synergy between PDT and PTT has led to its development as a combination therapy that has achieved impressive results compared to the use of PDT or PTT alone. Several groups have demonstrated that complete tumor ablation can be achieved using a PDT/PTT combined therapeutic method [431]. The incorporation of photosensitizers onto the surface of a photo-thermal nanomaterial to produce a single particle composite [76,431–439] is the most common strategy used to achieve a photo-thermal/photo-dynamic combination therapy. Since NIR-based nanoparticles can be engineered to efficiently interact with NIR radiation, tracking and detecting these agents during therapeutic treatments is possible at a deep tissue level. This optical imaging-guided approach and the combination of PDT and PTT phototherapy has gained substantial attention and has become a prosperous field for meeting clinical needs without the need for adverse surgery, chemotherapy, or radiation treatment.

Wang et al. have designed a new nanohybrid of rose-bengal (RB)-conjugated Au nanorods for use as an efficient in vivo photo-dynamic and photo-thermal treatment for oral cancer [440]. Although rose bengal (RB) is a well-known photosensitizer that generates singlet oxygen species with a high quantum yield (\sim 76%), the excitation wavelength is limited to 532nm light irradiation and fails to produce NIR fluorescence [441,442]. However in 2013, Tae and co-workers synthesized NIR nanogels embedded with Au nanorods (GNRs) and Chlorin e6 (Ce6) for in vitro and in vivo photo-toxicity applications [431] (Fig. 10A and B). Before photon treatment, effective tumor accumulation in vivo was detectable using the red-NIR fluorescence arising from the Ce6 lumiphore within the nanogel. Compared to that of independent treatments with PDT or PTT alone, a better anti-neoplastic effect was observed with PDT followed by PTT. A similar study on a hybrid of iron oxide@Au and methylene blue (MB) was performed by Ray and co-workers [432], where the fluorescence imaging could be captured using the MB molecule (with excitation at 650 nm and fluorescence signal detection between 680 and 720 nm). Photoablation of HaCaT cancer cells has also been achieved through a synergistic combination of photo-thermal and photodynamic treatments. Using an alternative method, where Au vesicles are used to gel-encapsulate Ce6 PS, the SPR coupling effect in the Au nanoparticle monolayer resulted in NIR absorption peaks and thus enhanced the NIR PDT and photo-thermal tumor treatment (671 nm and 808 nm laser at 2.0 W/cm²) (Fig. 10C). The accumulation within the tumor can also be tracked using a photoacoustic modality [443]. Simple assembly of the poly (dopamine) (PDA) nanoparticle followed by conjugation with Ce6 produced an excellent phototoxic PTT-PDT effect, with a combination laser irradiation of 670 and 808 nm and an



Fig. 10. Design of NIR excited PDT/PTT agents. (A) Scheme illustrating the procedure for loading the photo-agents (Ce6 and GNRs) into a Pluronic nanogel. (B) *In vivo* NIR fluorescence images of nude mice bearing SCC7 tumors after *i.v.* injection of photo-agents [431]. (C) Photosensitizer (Ce6)-loaded plasmonic gold vesicles (GVs) with trimodal (fluorescence/thermal/photoacoustic) imaging for use in photothermal/photodynamic cancer therapy [443]. (D) Schematic view of the preparation of PDA-Ce6 nanospheres for PDT and PTT treatments of HepG2 tumor-bearing nude mice. (E) *In vivo* NIR fluorescent images of HepG2 tumor-bearing nude mice 24 h after injection. (F) thermo-graphic images and 3D temperature distribution in tumor-bearing nude mice exposed to laser irradiation at 808 nm at different time points [433]. (G) A treatment scheme for the use of plasmonic copper sulfide (Cu_{2-x}S) nanocrystals (NCs) exhibiting both PTT and PDT capabilities [444]. Reproduced with permission [431,433,443,444].

extremely low dark toxicity [433] (Fig. 10D and E). This combined photo-therapy had high therapeutic efficiency both *in vitro* and *in vivo* compared with any single laser irradiation alone.

A non-invasive NIR therapeutic technique using combination therapy was also developed with graphene oxide as the PTT substrate followed by absorption of methylene blue as a potential photosensitizer [445]. Conjugation of Cy5.5 on the surface of the graphene oxide/MB was performed to track internalization of the particles under NIR imaging. By altering the PTT nanocore or the PS load, the following could be fabricated: gold-nanorod-PS layer-by-layer [446], sinoporphyrin sodium loaded graphene oxide [434,435], a pH-sensitive peptide inserting gold nanorod-photosensitizer conjugate [436], self-aggregation of Ce6 photo-sensitizers and gold nanorods [431,447], WS₂@BSA/MB nanosheets [437], and a ZnPc photosensitizer in a liposomal membrane decorated with a gold nanofilm [438]. Fluorescence, the PA technology, and an IR thermal camera with NIR light excitation, have been used for acquisition of both *in vitro* and *in vivo* images.

In a different approach, Vijayaraghavan et al. developed a new gold nanoechinus structure that was capable of NIR lightactivated dual modal photodynamic (NIR-I and NIR-II biological window) and photo-thermal therapy (808 nm) [389]. Interestingly, the photon energy can also be converted to emit excitation wavelength-dependent fluorescence for quantification of cellular uptake, as well as quantification of cellular markers, *in vivo*. In this case, the Au nanoechinus acted as both the photon-to-thermal converting agent and the photosensitizer without the need for addition of an organic dye to produce singlet oxygen. In another enlightened study, Xu and co-workers reported the optical photon-physical and photon-chemical properties of Au nanorods [448]. Au nanorods are among the most commonly studied photo-thermal therapeutic agents that use NIR laser excitation [449]. The authors demonstrated that singlet oxygen from Au nanorods could be generated with oneor two-photon excitation. The application of two-photon excitation at 808 nm resulted in a high quantum yield of singlet oxygen, when compared with rose bengal and ICG [450,451] and this difference was ascribed to the large two-photon absorption cross-section of the molecules in the Au nanorods.

Carbon nano-dots [452], quantum dots [453,454], silica/fluorescent donors [229] and Ln-based UCNPs [455] have also been demonstrated to be excellent nano-converters for the two-photon excited energy transfer to PS on PDT treatment. These results have been widely reviewed elsewhere [456].

The current discovery of total inorganic composite nanomaterials has allowed for the dual PDT/PTT capability to be obtained with plasmonic copper sulfide ($Cu_{2-x}S$) nanocrystals [444], WO_3 -x nanoparticles [457], Cd-based QDs [458–460], grapheme QDs [461–463], and Au nanorods [448]. In contrast, there are several reports on the use of the PDT/PTT platform

with organic nanocomposites, e.g., an NIR polymer and PS [433], IR825 and Ce6 in a nano-micelle [464], a human serum albumin-ICG hybrid [174], and doxorubin/ICG loaded lipid polymer nanoparticles [173]. Both infrared thermographic and NIR fluorescent maps are able to be captured and detected during the combined photo-therapeutic treatment *in situ*.

6. Nanotoxicology: concerns about the biosafety of NIR nanomaterials

The increasing number of studies and the production of numerous types of nanomaterials raises fundamental questions on their safety for use in humans and also for the environment. While classical toxicology has developed well-established tools and standards to assess the toxicity of various substances, their application to nanomaterial toxicity is problematic because these a new class of materials. Perhaps one of the most important aspects that deserves consideration is that nanomaterials have completely different physicochemical properties compared to the bulk form of the same material. Consequently, many aspects need to be taken into consideration when analyzing the impact of such materials on living organisms e.g. the fact that nanoparticles of the same material, and having the same shape but with a slightly different size may interact differently within a body thereby producing a different toxicity profile [465]. Parameters commonly accepted to be important determinants of nanoparticles toxicity are: structure, chemical composition, shape, surface composition, charge and area, redox properties, aggregation tendency, nature of the shell or coating material, chemical stability, biodegradability, as well as others [466]. Despite the growing number of studies, the mechanistic understanding of nanoparticles toxicity is still in its infancy.

With advances in the production of engineered nanomaterials and their broader applications in many aspects of life, exposure to nanoparticles has become an increasingly growing concern. Nanoparticles are extensively used in manufacturing; for example in new composite materials, protective coatings, inks, electronics, cleaning and disinfectant products, medicines, cosmetics and many other products. As a consequence, the penetration of manmade nanoparticles into the environment is inevitable.

Therefore, there is an urgent need for the introduction of reliable nanotoxicology methods that will allow for an assessment of the impact of nanoparticles on life in broad terms, including, toxicity at all levels of complexity (single cell, tissues, whole organisms), the impact on aquatic organisms, and the impact on various ecosystems. Exhaustive database on toxicology are a key part for the proper life cycle assessment of nanoparticles as products to be manufactured and used in a massive scale. This aspect, based on community expectations, should keep up with the technological progress and market needs [467].

Living organisms have evolved a number of adaptations to nanoparticles that naturally occur in the environment, however newly-fabricated materials present unique properties to which organisms have yet to adapt, and therefore they may become a serious health challenge.

Many studies have been conducted on the health effects of exposure to nanoparticles present as common pollutants, including airborne pollutants arising from the burning of carbon fuels and natural materials.

Ongoing discussions raise questions concerning their associated health hazards, the balance between benefits and threats, and on safe handling procedures. The Organization for Economic Cooperation and Development (OECD) has recently suggested a critical revision of currently employed methods for analysis of materials safety to specifically address man-made nanoparticles [468].

The implementation of new analytical methods may highlight so far undiscovered facts concerning the property of nanomaterials. For instance, some studies have shown that titanium dioxide and zinc oxide nanoparticles, although already recognized as safe, can under certain specific conditions be toxic to the human brain and lung cells [469–471].

6.1. Methods to analyze the toxicology of NIR nanomaterials

The number of available research models and methods for the nanotoxicity is constantly growing, enriching our database on the impact of nanomaterials in living systems.

The most commonly applied approach for the initial estimation of potential nanoparticle toxicity is based on *in vitro* studies using various cell lines. Living cells in a cell culture are treated with test materials and after a certain time (usually from a few to 24–48 h) their viability, morphology, ability to proliferate, as well as other parameters are analyzed. Viability and morphology can be studied directly using microcopy, or can be estimated in a more quantitative way using one of the many viability assays that are currently available on the market (MTT, MTS, XTT, WST-1, CCK-8, Resazurin viability assay, etc.). In general, their principle mechanism of action is the formation of colored or fluorescent compounds in the presence of active mitochondrial enzymes [472–474]. Other popular assays include, the lactate dehydrogenase (LDH) release assay, a test indicative of an effect on cell membrane integrity associated with the release of the cytosolic enzyme LDH from cells [475]; tests for ROS (reactive oxygen species) that allow analysis of the formation of ROS as a result of oxidative stress [476]; measurement of apoptosis markers (for e.g. the Annexin V test) that are used to observe whether the tested material has the ability to induce programmed cell death [477]. More complex studies can include immunohistochemistry or analysis of inflammatory cytokines released by cells treated with substances having pro-inflammatory activities. These types of tests involve ELISAs, quantitative real-time PCR, or flow cytometry [478].

The proliferation potential of cells can be easily estimated via thymidine incorporation assays (e.g. the older radionuclide incorporation assay or modern methods developed for flow cytometers) [479].

Information on the impact of nanomaterials on cellular genetic material (genotoxicity) can be gained using the simple and sensitive Comet Assay. In this assay DNA obtained from lysed cells embedded in agarose is subjected to electrophoresis at high pH resulting in structures resembling a comet when observed under a fluorescent microscope. The extent of the "comet tail" reflects the number of DNA breaks. This method allows for analysis of DNA breakage at a single cell level [480].

In vitro techniques are usually the methods of choice for most researchers during "proof of concept" or initial toxicology studies, and they can provide valuable data with relatively low cost and effort, and the necessary resources are usually available in standard research laboratories.

However, it must be considered that the data obtained from *in vitro* studies carries a significant level of uncertainty due to many constraints. First of all, most studies are performed on models, specifically cell lines, that are generally transformed, immortalized cells, and thus their function and responses to compounds may differ to some extent from normal cells. Normal cells have for example limited proliferative potential in comparison to cell lines. Cells in culture dishes function in an isolated surrounding; all of the complexity of interactions between cells in tissues, organs, and the whole body are missed from this analysis meaning that the results obtained may differ significantly from the real-life situation. To support this notion, one example can be given: it is well know that bacterial endotoxin (LPS) exerts it full toxic potential at the whole body level whereas at the cellular level cytotoxicity is lower and is restricted to certain cell types [481,482].

The potency of LPS relates to its ability to strongly stimulate the release of inflammatory cytokines by immune cells, overstimulation leads to the damage of vital organs and death, therefore an indirect mechanism is responsible for the toxic effect of LPS at the whole body level. Similarly, a lack of noticeable cytotoxicity of a nanomaterial in *in vitro* studies may not necessarily reflect the same at the whole organism level. On the other hand, it is much more probable that a compound demonstrating toxic effects in *in vitro* models will be toxic when applied to a body than vice versa.

New models for *in vitro* studies are continually being sought, e.g. pluripotent stem cells [483], in order to improve the predictive value of the data obtained.

Other difficulties noticed by many researchers are that interactions of certain new nanomaterials with components of assays can significantly affect measured values, as has been specifically noted for carbon-based nanomaterials [206]. The use of novel instrumentation that utilize microelectronic sensing devices capable of assessing cell viability and as well as other cell parameters may help to resolve the issue [484].

Toxicity studies conducted in models with higher levels of complexity, typically *in vivo* models, will cover many other parameters and can examine the effects of mutual complex interactions between diverse types of cells, tissues and organs within a body. The models provide more comprehensive toxicological data, but at significantly higher costs, and the technical complexity of the research, including ethical issues that are associated with animal studies are an additional burden.

Although there are several toxicology models based on invertebrate organisms, including *Caenorhabditis elegans*, *Droso-phila melanogaster* [485] and others, the most popular models remain laboratory rodents (mice, rats, rabbits). Study on these common laboratory animals allow the investigation of more aspects of toxicity including bio-distribution, accumulation, excretion, and metabolisms following changes in dose and routes of administration.

Recent reports have also proposed the use of zebrafish as a versatile animal model to assess nanoparticle toxicity in a broad range of aspects including: acute toxicity, immunotoxicity, genotoxicity and gene expression, neurotoxicity, impact on fertility, etc. Significant advantages of this model include ease of handling, small size, high reproducibility, fast development and transparency of the embryo.

Different research methodologies are used to study toxicity in animal models. Following administration of a test substance, animals are monitored for symptoms of toxicity and adverse effects including death, diarrhea, lethargy, depression, suppression of movement, skin and eye irritation, swelling, depressed water and food uptake, and behavioral changes [486].

Depending on the study design, many biochemical and physical parameter can be tested using blood, urine, and feces sampling. Commonly, these biochemical markers, as well as hematological blood parameters (e.g. WBC – white blood cells count) are used as markers of health status. Furthermore, liver and kidney function can be monitored indirectly, as those organs play key roles in detoxification, excretion, and accumulation of foreign materials. In the case of an end point study, animals are euthanized and subjected to histological analysis of tissues and organs. Studies examining the effects of chronic exposure at low doses, impacts on genetic material, and fertility impact usually require long-term experiments and careful experimental design.

Recent reports have presented new interesting ideas to study nanotoxicity that may open new possibilities for the field. For example, Ivask et al. demonstrated a genomic approach to assess the different toxicities of different forms of a nanomaterial [487].

There have also been attempts to develop computational methodologies for the analysis of quantitative structure-activity relationships (QSARs). This technique might help to predict toxicity of nanoparticles based on correlation with specific parameters like size, shapes, coating, and porosity [488].

6.2. Toxicity of NIR nanomaterials

Several comprehensive review articles discussing the toxicity of engineered nanoparticles have been recently published, where quantum dots, gold, iron oxide nanoparticles [489], or lanthanide doped nanoparticles [490] were compared, and the influences of nanoparticle shape, size, and surface functionalization on cellular uptake were described [491,492].

Data on the toxicity of NIR materials are still scarce because these materials are relatively new to the field so there has been insufficient time to obtain enough data for conclusions to be drawn. Certainly, a long-term and systematic study is needed to draw reliable conclusions concerning the safety of these materials. So far, some features can be anticipated, based on comparison to known characteristics of other nanoparticles formed from analogous bulk materials.

An important notion is that most elements or materials employed for the formation of NIR nanoparticles (see Table 1) are not toxic in the bulk form, therefore eventual toxicity may be attributed to the nanoparticle form. Apparent toxicity applies to such elements like cadmium ions (in Cd based QDs), silver, other noble metals, and fluoride ions (in UCNPs).

NIR materials based on Au are probably the best characterized in terms of physical properties and biocompatibility [5,493]. In general, most spherical forms of AuNPs show little or no toxicity apart from particles with some specific sizes e.g. a nanoparticle with diameter of 1.4 nm that fits into the DNA major groove had increased toxicity [494]. It should be remembered that the surface coating strongly affects the properties of AuNPs including their bio-distribution and toxicity. Gold-based NIR materials are mainly represented by: nanorods, nanoshells and nanocages. Gold nanorods exert slightly higher toxicity than spherical particles; but this can be reduced by replacing hexadecyltrimethylammonium bromide (CTAB), a surfactant necessary for the synthesis of nanorods, with phosphatidylcholine, for example [495]. Gold nanoshells and nanocages, have been studied less extensively but so far no serious toxicity problems have been observed [331,496].

Noble metal NIR materials such as Pd nanosheets (<10 nm diameter), when formulated with glutathione, showed prolonged blood circulation, good accumulation in tumors, with no apparent toxicity in mice and were renally cleared [497]. Other recently developed materials such as WS₂ nanosheets, MoS₂ nanosheets also showed good stability and biocompatibility [498–500].

However, some intrinsic toxicity was observed for CuTe nanoparticles. However, further work using better encapsulation or more strictly controlled administration were suggested as being effective at addressing this problem [501].

Iron oxide nanoparticles possesses unique and useful properties and have been widely used in the biomedical field, as they are considered as safe and biocompatible [502]. Polypyrrole (PPY) NPs are another recently developed material with very promising features including photostability and a low synthesis cost. They have been studied for suitability as a photo-thermal cancer therapy with outstanding results both *in vitro* and *in vivo* [503,504].

Carbon-based nanomaterials have recently gained significant interest due to their spectacular properties and potential application in many fields. As of now, they will certainly take a leading position among the nanomaterials produced in high quantity. However, possible environment release and chronic exposure create safety concerns. A constantly growing number of reports predict that their application in nanomedicine will also be substantial. Studies *in vitro* and *in vivo* models have shown that some forms of graphene family nanoparticles showed toxic properties that were strongly dependent on dose, size, surface shape, chemistry, etc.; these reports are extensively discussed elsewhere [505].

Given that production and application of nanomaterials is a relatively recent achievement, nanotoxicology is therefore an emerging field and will have to solve problems that so far have not been encountered in classical toxicology. Nanoparticles have may complex features including size, shape, charge, coating and changes in these features may produce very different toxic effects. In addition, route of administration and factors such nanoparticle degradation, metabolism, adsorption to other molecules, and adsorption to tissues may all affect their toxicity profile.

These factors therefore create an enormous number of variables that have to be considered and analyzed during toxicology studies. The question that is raised is how to establish reliable standardized protocols for such studies. Although this question remains to be answered, it is widely recognized across the scientific community.

The evaluation of the safety of nanoparticle-based drugs or medical products also has the problem of extrapolating from *in vitro* and *in vivo* studies to the real-world clinical situation. For example, the doses used in toxicity studies are usually much higher than those generally required in routine clinical use as diagnostics or treatments, so the available data may not be relevant. Another problem is that most data is generated using short-term exposure experiments whereas long-term studies are necessary to really assess the effect of chronic exposure to NPs, including the effects of NPs that have accumulated in organs over time.

In conclusion, the existing data allow us to propose several new NIR materials, with no apparent toxicity, as candidates for medical applications. However, a final conclusion on biosafety will be possible only after thorough, detailed, studies (including clinical stage) have been conducted, and these data will have to apply to a particular product than a class of nanomaterials.

6.3. Regulatory issues on nanomaterials

Despite the fact that nanoparticle properties are rarely seen in their bulk counterparts and show promise in many alternative and new bio-medical applications, such as bio-sensing and bio-imaging, the toxic effect is very difficult to predict based solely on chemical composition. This issue originates from a number of factors, including surface chemistry (mostly related to charge) and surface to volume ratio, nanoparticle size, the nanoparticle's shape anisotropy, chemical composition, and susceptibility to dissolution in organic media, which all have been shown to affect their cytotoxicity to some degree. In general, nanomaterials are more chemically active and behave in different ways compared to their bulk component chemicals. They are also more likely to be taken up by humans and animals, through the skin, and lungs, and through food ingestion [490] and may then accumulate in different organs (liver, spleen, lungs, etc.). Moreover, the quantification of nanotoxicity at the cellular level *in vitro*, is not relevant for extrapolation of *in vivo* nanotoxicity. A simple example, which

Table 1

Overview of the toxicity aspects associated with NIR nanomaterials.

Material	Coating	Size	Research model	Dose/concentration	Treatment time	Results	Ref.
Metallic and metallic co	ompound-based nanoparticl	les					
Au nanorods	Chitosan oligosaccharide	47 ± 3 nm/11 ± 1.8 nm	Oral adenosquamous carcinoma cell line CAL 27	1–200 μg/mL	72 h	Viability: 90–100%	[329]
	lactate/chitosan anti anti-EGFR Ab		Male 6-week-old BALB/c nude mice	100 µg/i.v. injection	7 days	No significant changes in histology, hematological, and clinical biochemistry parameters apart from slight reduction of RBC count	
Au nanorods	PEG	65 ± 5 nm/11 ± 1 nm	HeLa cells	0.5 mM/Au atom	24 h	Viability: 90%	[330]
Silica core Au nanoshells	SH-PEG	55-nm core radius and a 10-nm-thick	Human breast epithelial carcinoma SK-BR-3 cells	4.4×10^9 particles per mL	3 h	No cell death (microscopy observation)	[331]
Porous hollow Au nanoparticles	Bare or SH-PEG coated	150 nm	Prostate cancer cell line (PC- 3), human breast carcinoma cell line (MDA-MB-231), lung-metastasized prostate cancer cells (PC-3ML)	0.8, 4, 20, and 100 μM	24 and 72 h	Both types non-toxic in all concentrations, observed some growth inhibition of PC-3, PNT1A, and MDAMB231 cells at higher concentrations, and growth enhancement in PC3-ML cells after 72 h	[332]
Silica core Au nanoshells	-	30 nm core/shell thickness of 5–10 nm	Prostate cancer cell lines, LNCaP and PC3	$4\times 10^{12} \text{ particles/mL}$	24 h	No toxicity detected (MTT assay)	[333]
Silica core Au nanoshells	Anti-HER2 or a nonspecific antibody (anti-IgG)	120 nm core/10 nm shell	SKBr3 breast adenocarcinoma cells	$3 imes 10^9$ nanoshells/mL	From 1 h	No toxicity detected (calcein fluorescence)	[334]
Au nanocages	PEG-anti-EGFR Ab or PEG	50 ± 3 nm and wall thickness of 5 ± 1.2 nm	U87MGwtEGFR cells	0.02 nM	Up to 24 h	No toxicity observed (microscopy)	[335]
Au nanocages	Peg 5000	33 nm in edge length	Mice bearing EMT-6 tumors	$1.7\times 10^{12} \text{ particle/mouse}$	1, 6, and 24 h	No signs of toxicity observed (behavioral observations)	[336]
Pd nanosheets	Reduced glutathione (GSH)	4.4 nm	Mice	400 μg/mouse	40 days	No sign of toxic side effects within 40 days. Neither death nor significant body weight drop. No noticeable signal of organ damage	[497]
Pd nanosheets	Doxorubicne/ reduced glutathione	4.4 nm	Mice	300 μg/mouse	40 days	No changes in major organs (histological examinations)	[337]
Pd nanosheets	No coating	16 nm	Immunodeficient nude mice	160 μg/mouse	7 days	No changes in histological image of major organs	[338]
WS ₂ nanosheets	PEG coated and no coated	1.1 nm thickness/50– 100 nm diameter	Murine breast cancer cells (4T1), HeLa, human embryo kidney cells (293T)	0.1 mg/mL	24 h	No toxicity detected (MTT assay) for PEGylated particles, non-PEGylated were toxic at high conc.	[498]
WS ₂ nanosheets	BSA	Average thickness 1.6 nm/diameters of 20–100 nm	HeLa	Up to 0.2 mg/mL	24 h	Viability: >80%	[437]
MoS ₂ nanosheets	PEG, loaded with chlorin e6 (Ce6)	Thickness $\sim 1 \text{ nm}$	Mice bearing 4T1 tumors	6.85 mg/kg	20 days	No significant side effects in major organs (histological study)	[499]
MoS ₂ nanosheets	PEG	Thickness 0.29 nm/80 diameter; thickness 0.39 nm/103 nm dimeter	4T1 cells L929 cells	200 μg/mL	24 h	No significant cytotoxicity was observed (CCK-8 assay, morphology)	[500]
			BALB/c nude mice	200 μg/mouse	Up to 40 days	No changes observed in major organs	

Table 1 (continued)

Material	Coating	Size	Research model	Dose/concentration	Treatment time	Results	Ref.
<u> </u>	DEC			500 14	261	(histological examination)	15001
Cus	PEG	l l nm	U87 glioblastoma cells	500 μΜ	26 h	No apparent change in cell viability (microscopy)	[506]
Graphene oxide/CuS nanocomposite	PEG	13 nm	HeLa cells	500 μg/mL	24 h	Viability >90% (MTT assay)	[507]
FeS nanoplates	PEG	32–36 nm	BALB/c mice	100 mg/kg	1, 7 and 50 days	No changes in biochemical parameters of blood, kidney, liver, no changes observed in major organs (histological examination)	[508]
Ag ₂ S QDs	PEG	5.4 nm	BALB/c mice	15 and 30 mg/kg	Up to 60 days	No interference on the mice growth, blood biochemistry not changed after 60 days, transiently elevated AST value, drop in platelet and WBC counts at the first few days (normal level after 14 days)	[328]
ZnS:Mn/ZnS core/shell nanoparticles	Mercaptopropionic acid/folic acid	8.0 nm	HeLa cells	0–400 mg/mL	6 h or 24 h	Viability: about 90% at highest dose	[509]
CuTe NP	Poly(isobutylene- alt-maleic anhydride)	10–20 nm	3T3 embryonic fibroblasts	75 nM	3 h	Certain toxicity was visible (DAPI labeling)	[501]
Cs_xWO_3 nanorod	PEG	Diameter ~11 nm, length ~50 nm	HeLa	0.5 mg/mL	24 h	Viability >90%	[392]
Tungsten oxide nanorods	PEG	Diameter of 4.4 ± 1.5 nm, length of 13.1 ± 3.6 nm	HeLa cells L929 cells	Up to 1000 µg/mL	12 or 24 h	Viability – 80% up to conc. 500 µg/mL, decrease in viability at higher concentrations	[400]
W ₁₈ O ₄₉ nanowires	PEG	Length of 80–400 nm, thickness 0.9 nm	HeLa cells	0.25–3.0 mg/mL	24 h	No significant differences in the cell proliferation, viability greater than 90% (MTT assay)	[393]
W ₁₈ O ₄₉ nanoparticles	Anti-HER-2 monoclonal antibody	4.5 nm mean diameter	Human alveolar basal epithelial cell line A549	Up to 5 mg/mL	28 and 52 h	Decrease 16.8% in viability for the highest concentration 5 mg/mL	[394]
Na _{0.3} WO ₃ nanorods	PEG	Diameter ~5 nm, length 39 nm	TC71 tumor cells	Up to 1 mg/mL	24 h	Viability 98.9% at 0.5 mg/mL and over 80% for 1.0 mg/mL	[510]
BaTiO ₃ nanoparticles	Glycol-chitosan	285 nm	Human neuroblastoma SH- SY5Y cell line	0, 5, 10, 20, 50 and 100 μg/mL	48–52 h	Viability 90% (MTT assay) no membrane damage (live/dead viability/cytotoxicity test), absence of apoptosis (Annexin V test), not detectible oxidative stress signs	[511]
KTiOPO4 nanocrystals	Bare	80 nm	Cortical neurons from mouse embryonic brain	Not reported	30 min exposure to NP, 2–4 days culture	Dendritic growth of cortical neurons not affected	[512]
CdTe QDs CdTe/CdS CdTe/CdS/ZnS core- shell-shell quantum dots	Bare	2.15 ± 0.26 nm 3.01 ± 0.42 nm 4.22 ± 0.52 nm (core sizes), 3.01 and 4.22 nm for core/shell NPs	HeLa	Up to 300 nM	24 h	CdTe-cytotoxic (even at lower conc. – 75 nM), core-shell CdTe/CdS QDs less toxic, viability ~80% at conc. 300 nM, lowest toxicity for CdTe/CdS/ZnS QDs – ~100% at conc. 300 nM	[513]

 Table 1 (continued)

Material	Coating	Size	Research model	Dose/concentration	Treatment time	Results	Ref.
Nanodiamonds	-	2–10 nm	Kun Ming mice	0.8 mg/kg, 4 mg/kg, 20 mg/kg body weight (intra-tracheal instillation)	3 days	Lung toxicity observed (histopathological examination), changes in biochemical parameters of BAL fluid, kidney, liver and blood	[514]
	CM-dextran or BSA	~120 nm	Caenorhabditis elegans	1 mg/mL	3 h up to a few days	No change in longevity and reproductive potential, no symptoms of detectable stress to the organism	[515]
Single-walled carbon nanotubes	Mouse (anti-human CD22) IgG	Length 0.2–1.4 μm (average of 0.59 μm)	Burkitt's lymphoma cell line (Daudi cells)	3.6 µg	24 h	No toxicity observed ([³ H]thymidine incorporation assay)	[516]
Single-walled carbon nanotubes	PEG (covalently bound)	300 nm to a micrometer	Male KM mice, BALB/c mice bearing EMF6 tumor and C57BL mice bearing Lewis tumor	2.4 mg/kg body weight	1 h, 1 day, 3 days and 7 days	No signs of acute toxicological responses or clinical abnormalities	[517]
Single-walled carbon nanotubes	Bare	Diameter of ca. 1.4 nm and a mean length of	Alveolar macrophages from guinea pigs	1.41, 2.82, 5.65, 11.30, 28.20, 56.50, 113.00, and 226.00 $\mu g/cm^2$	6 h	Significant, dose dependent toxicity	[518]
Graphene QDs	PEG	3–5 nm	HeLa, A549	10, 20, 40, 80 and 160, 320, and 640 μg/mL	24 h	95% viability at conc. 160 µg/mL, 85% at 640 µg/mL, negligible apoptosis or necrosis, no signs of oxidative stress or membrane damage	[519]
			BALB/c mice	20 mg/kg (7 injections)	40 days	No difference in major organs in comparison to controls, hematological markers not changed apart lower WBC (however still within a normal range), blood biochemical markers not changed	
Carbon nanodots	Bare PEI PEG	4–7 nm	NIH/3T3	5–400 μg/mL	24 h	Dose and charge dependent toxicity, negatively charged (PEI) highest tox. $IC50 \sim 5 \mu g/mL$ low tox. for PEG coated CDs	[520]
Carbon nanodots	-	2–6 nm	CHHO-K1 COS-7 HeLa	0–10 mg/mL	24 h	No apparent toxicity at dose up to 5 mg/mL, dose dependent, low tox. at higher conc.	[521]
Carbon nanodots Graphene oxide nanosheets and reduced graphene oxide nanosheets	– PEG	2−6 nm Mean of ~18.8 nm	HepG2 cells U87MG, MCF-7 human epithelial breast cancer cells	0.1–1 mg/mL Up to 1 mg/mL	24 h 48 h	Cell viability 90–100% (MTT assay) Dose dependent toxicity observed: IC_{50} of ~80 µg/mL (nano-rGO) and IC_{50} of ~99 µg/mL for nano-GO	[522] [523]
Graphene nanoplates intercalated with manganese	-	Diameter – 200 nm and 3 nm thickness,	NIH3T3 mouse fibroblasts, A498 (human kidney epithelial cells)	1–500 μg/mL	24 and 48 h	Dose dependent toxicity observed: IC ₅₀ range 179–301 µg/ml (LDH and calcein-AM assay)	[524]
Other materials Mn-doped Si QDs	Dextran or dextran sulfate	4.3 ± 1.0 nm	P388D1 cells, mouse embryonic fibroblast NIH 3T3 cells	0.65, 1.31, 2.61, 5.22, and 10.45 mg/well and 0.64, 1.27, 2.55, 5, 10, and 10,20 mg/well	24 h	Resazurin viability assay viability – 90% (P388D1 cells) and 97% (NIH 3T3 cells)	[525]
Polypyrrole nanoparticles	-	Average size of $\sim 50 \pm 5$ nm	BALB/c mice	10 mg/kg	60 days	No changes observed in major organs (histological examination)	[503]
Polypyrrole nanoparticles	-	46 nm	Hela cells	300 μg/mL	12 h	Viability ~90%	[504]

demonstrates the level of complexity in nanotoxicity studies relates to the definition of dose of nanoparticle-based compounds. Should the mass of the NPs, their total number, or the surface area per unit volume be considered as the dose? These issues and question are in stark contrast to definition of toxicity based purely on chemical composition, making nanotoxicity studies using conventional methods unreliable and guestionable. The European Commission's Scientific Committee on Emerging and Newly Identified Health Risks (SCENIHR) noted that "experts are of the unanimous opinion that the adverse effects of nanoparticles cannot be predicted (or derived) from the known toxicity of material of macroscopic size, which obey the laws of classical physics" (please refer to the European Commission's Scientific Committee on Emerging and Newly Identified Health Risks (SCENIHR) for a more detailed discussion concerning the appropriateness of existing methodologies to assess the potential risks associated with nanotechnologies). Similarly, the U.K. Royal Society and the Royal Academy of Engineering emphasized, "Free particles in the nanometre size range do raise health, environmental, and safety concerns, and their toxicology cannot be inferred from that of particles of the same chemical at a larger size" (please see the report from Royal Society of Engineering on 'Nanoscience and nanotechnologies: Opportunities and uncertainties'). Furthermore, the Institute of Occupational Medicine notes, "Because of their size and the ways they are used, they have specific physicalchemical properties and therefore may behave differently from their parent materials when released and interact differently with living systems. It is accepted, therefore, that it is not possible to infer the safety of nanomaterials by using information derived from the bulk parent material" [526].

There are some arguments against developing specialized regulations for nanotechnology products. This is because the existing methods exploit the most advanced available scientific methodologies to assess risks and safety, and they have been successful in identifying dangerous or unacceptable materials or products. Rather than creating a dedicated regulation based on size, it is postulated to treat particle size as one of the several parameters which define a substance to be approved. Another argument against global regulation originates from the fact that nanotechnology applications are often hypothetical with the strongest impact far in the future, thus regulating such futuristic technologies is vague. At the same time, the nanomaterials and products based on them, are not much more challenging than any other new materials. Therefore, a minor tuning to already existing regulatory schemes is faster and more feasible than introducing global regulations [527].

The regulatory agencies of the European Union, the United States and Australia agree on the accurateness of existing regulations with respect to nanotoxicity, however numerous initiatives are being considered at the national and international level to decide whether additional studies are necessary. Some preliminary approaches to understand the role, and potential pitfalls of nanotechnology, were initiated in Europe in 2004, when a warning about the necessity of addressing any potential negative impacts of nanoparticles on public health, safety, and the environment were articulated (please see the report from the Commission of the European Communities, Communication from The Commission, Towards a European strategy for Nanotechnology, 2004). The expected future impact of nanotechnology on the quality of life, materials sciences, healthcare, information technology, and the environment has been acknowledged much earlier by many countries including the USA, Japan, Europe, China, and Russia. At that early time, the potential advantages of nanotechnology dominated the potential risks associated with the extensive production and use of nanomaterials. Eight years later, in 2011, the European Commission announced and adopted a definition of nanomaterials [528]. Existing knowledge concerning nanomaterials was summarized in 2012 in the Commission Communication on the Second Regulatory Review. At that time, carbon black and amorphous silica were the most predominant nanomaterials in the market-place but new nanomaterials, such as nanotitanium dioxide, nano-zinc oxide, fullerenes, carbon nanotubes, and nanosilver were fast gaining interest for use in new applications such as in catalysts, electronics, solar panels, batteries and in the biomedical field. The communication concluded (please see communication from the Commission to The European Parliament, The Council And The European Economic And Social Committee, Second Regulatory Review on Nanomaterials, 2012), that although nanomaterials "are similar to normal chemicals/substances in that, some may be toxic and some may not", possible risks are actually related to a given nanomaterial for a specific application, therefore the risk assessment shall be examined on a case-by-case basis. Despite the fact that current toxicity assessment methods have been found applicable to nanomaterials, continuing work on particular aspects of risk assessment was expected and the EU Commission advised modification of the Registration, Evaluation, Authorisation and restriction of CHemicals (REACH) Annexes and further development of guidance for registration of new (nanotechnology based) chemicals after 2013. Similar to REACH, the Scientific Committee on Emerging and Newly Identified Health Risks (SCENIHR) stated that while the existing toxicological and eco-toxicological methods are applicable to quantify and estimate many of the threats linked to the production and exploitation of nanoparticles, these methods may not be sufficient to address all of the hazards (please see Scientific Committee on Emerging And Newly Identified Health Risks (SCENIHR), Opinion on The appropriateness of existing methodologies to assess the potential risks associated with engineered and adventitious products of nanotechnologies, 2005). It should be noted, that many European countries (the German Federal Institute for Risk Assessment; the Department for Environment, Food and Rural Affairs (DEFRA) in the United Kingdom; the French Ministry of Ecology, Sustainable Development and Energy; the Danish Consumer Council and the Danish Ecological Council in cooperation with Technical University of Demark; the Netherlands National Institute for Public Health and the Environment-RIVM, as well as others) have developed their own regulations and guidelines, which more precisely described how approvals, labeling, reporting, or communication with REACH shall be performed when it comes to new products which involve nanomaterials [529].

In the USA, the advantages and industrial applications of nanomaterials have also been exploited before risk assessment methods had been developed and nanotoxity had been understood. In 2007, the United States Food and Drug administration stated that it does not see the need to develop any regulatory definition of nanomaterials and nanotoxicology (please see the

Nanotechnology Task Force Report). However, in response to growing interest and commercial exploitation of nanomaterial products on the market, FDA decided to issue preliminary guidelines in 2011 (please see draft guidance: Considering Whether an FDA-Regulated Product Involves the Application of Nanotechnology) to indicate whether a product, which is a subject to FDA approval, contains any nanomaterials or involves nanotechnology products. The following year, more details were provided with respect to the food and cosmetic industries, which involved the use of nanomaterials (please see FDA Draft Guidance for Industry: Assessing the Effects of Significant Manufacturing Process Changes, including Emerging Technologies, on the Safety and Regulatory Status of Food Ingredients and Food Contact Substances, Including Food Ingredients that are Color Additives). Unlike in Europe, the FDA concluded in 2012, that safety assessment methods existing in the USA, are satisfactory for a variety of materials including nanomaterials. In 2001 a National Nanotechnology Initiative (NNI, www.nano.gov) was established as a response to the natural shift from the fundamental studies on the synthesis and characterization of nanomaterials towards exploiting nanotechnology-enabled products (e.g. in electronics, clean energy technologies, clothing and fabrics, car industry, the biomedical and drug market, etc.). Between 2009 and 2016, the US revenue from the sale of nanotechnology-based products increased over sixfold to reach ca. 500 billion US dollars. The most recent NNI Strategic Plan was issued in October 2016. The document discusses: (i) advancing a world-class nanotechnology R&D program, (ii) fostering the transfer of new nano-technologies into products for commercial and public products, (iii) developing and sustaining educational resources to advance nanotechnology and (iv) support responsible development of nanotechnologies. In the mean-time the US Environment Protection Agency (EPA) released numerous nanotechnology white papers (like EPA 100/B-07/001, February 2007), underlining the necessity to control and monitor the production of nanotechnology containing products. Recently the EPA has issued a final version of its rules, which oblige manufacturers to report manufactured or processed nanoparticles, which, as defined in Section 3 of Toxic Substances Control Act (TSCA), are solids at 25 °C and standard atmospheric pressure; are manufactured or processed in a form where any particles, including aggregates and agglomerates, are in the size range of 1–100 nm (nm) in at least one dimension; and are manufactured or processed to exhibit one or more unique and novel properties. This 'definition' of nanomaterial differs from the EU definition, which states that nanomaterial 'means a natural, incidental, or manufactured material containing particles, in an unbound state, or as an aggregate or as an agglomerate, where, for 50% or more of the particles in the size number distribution, one or more of the external dimensions is in the size range of 1–100 nm. The EPA rule does not apply to chemical substances manufactured or processed in forms that contain less than 1% by weight of any particles, including aggregates and agglomerates, in the size range of 1–100 nm. It is important to mention that according to the EPA, 'unique and novel properties' means the reportable chemical substances are not just compounds containing nanoparticles (size 1-100 nm), but they must also demonstrate a size-dependent property different from the properties of the material at sizes greater than 100 nm, which is actually the purpose that the chemical was manufactured or processed to have that form or size.

The situation in Australia is similar to that in the USA. A National Nanotechnology Strategy has been developed (please see the report from Australian Government, Approach to the Responsible Management of Nanotechnology). A 2008 review of Australia's regulatory framework (see a Review of Possible Impacts of Nanotechnology on Australia's Regulatory Framework Final Report September 2007) concluded that there were significant regulatory legal gaps that should be addressed. Even though it was obvious nanomaterials behave differently to bulk forms of the substance, most regulators (National Industrial Chemicals Notification and Assessment Scheme, Department of Environment or Therapeutic Goods Association) do not require a separate risk assessment of nanoforms of existing substances.

7. Current challenges and future perspectives

7.1. Sensitivity and background interference

The decreased light absorption of water, pigments, and fluorescent proteins provides an optimal signal-to-background ratio for NIR optical imaging, which makes it easy to capture/generate images and thermal signals up to a distance of a few millimeters [530]. However, non-negligible red to NIR auto-fluorescence from hemoglobin, melanin, lipids, and other endogenous fluorophores, may still be multi-photon excited by NIR femtosecond lasers [531]. Auto-fluorescence in the visible wavelengths is an important issue because it interferes with signal detection *in vivo* [532]. These auto-fluorescence signals will mask the desired signal and severely limit the target-to-background ratios [533]. For example, the commercially available Cy5.5 molecule has excitation/emission at 675 nm/694 nm which can be applied for molecular imaging in tracing. To avoid the absorption and scattering at short wavelengths and the elevated water absorption over 950 nm, imaging based on an NIR multi-photon process can yield the deepest tissue penetration, with improved resolution, the highest sensitivity, and with minimal photodamage/photobleaching (Fig. 11) [534]. Although NIR excitation can greatly suppress auto-fluorescence, it still has a major limitation in the lack of a suitable scale-up synthesis method for the production of biocompatible, large absorption cross-sections, high quantum efficiency, multiphoton, imaging nanoprobes.

Another strategy for improving sensitivity is by designing target-specific substances conjugated to NIR materials. The surface groups would help find the disease tissue and/or sense the targeting area. The substantial accumulation and internalization of desired particles would therefore be expected to locally increase signal generation. Additionally, in order to be an effective treatment at focused sites in solid tumors, the accumulation of the treatment in a very small area will prevent



Fig. 11. (A) Tissue auto-fluorescence is much lower at NIR wavelengths. An untreated nude mouse was imaged with visible (532 nm) and NIR (700 nm or 800 nm) light. Auto-fluorescence at 532 nm (the Cy3 channel) was very high [532]. (B) A schematic showing how a tumor can be imaged directly through the skin in a live animal. The optical imaging modality (that is, one- or three-photon) and properties of the probes will determine the imaging penetration depth and resolution. If multiple tissue components are labeled with different color dyes, it is possible to image them simultaneously [534]. Reproduced with permission [532,534].

injury to the surrounding tissue. Indeed, targeting substances using a small peptide, DNA or an antibody can increase the binding selectivity in many pre-clinical and clinical studies [535,536].

Current studies have moved to further understand the molecular mechanisms for the motion, apoptosis, and necrosis of living cells in response to light [401,537–539]. In addition to the effect on cell migration, these reports also describe a proinflammatory response whereby pro-inflammatory cytokines are released into the extracellular milieu. These results have educated researchers and doctors to be more careful, by using the appropriate laser wavelengths, power density, and drug doses, in procedures aimed at disease healing, restoration and ablation.

7.2. Technical hurdles and potential solutions

In the clinic, therapeutic and surgical lasers commonly use He-Ne, Nd:YAG, and CO_2 lasers in the red-to-IR wavelengths. These laser systems traditionally require high-cost pump lasers. However, the laser power supply pumps even more energy in a very short time into the living system, which results in the generation of hot plasma and evaporation and damages the healthy tissue in addition to the tissue of interest. Therefore, the operation power needs to be lower than 250–670 mW/cm² to avoid a thermal effect that would otherwise damage the tissues [540].

Among the f-block, d-block, defect-related nanoparticles and organic dye donor-acceptor pair hybrid, lanthanide-doped UCNPs are a major focus of current research aimed at achieving better spatial and higher resolution optical imaging. Their advantage is the utility of low power ($\sim 10^{-1}$ W/cm²) and the use of low cost CW laser diodes to generate upconverted

photons, which is in contrast with the high power density excitation (10^6 W/cm^2) needed with an expensive NIR ultrafast pulsed laser in nonlinear optical processes [140]. Early work focused on broad applications in biomedicine using different single solid nanocores and core-shell nanostructures with specific surface properties [84,140,321,541–543].

Size-dependent ODs have a multicolor tracking capacity and have been used to monitor multiple antibody labeling of cells in vitro and nanomedicine distribution in vivo. However, they suffer from toxicity to tissues in vivo. Therefore, there is an urgent need to replace the traditional toxic QD with new optical nano-reagents. A series of UCNPs with different Ln dopants had NIR-to-visible emission compatible with use as optical bio-probes. With adjustments in the Y/Yb/Er ratio, Ln-doped NaYF₄ UCNPs were created that exhibited different color emission peaks of blue, green, and red, as reported by Liu and co-workers [544]. These molecules provided a new insight into the multicolor tracking required for lymph node mapping. Compared to conventional QDs [82], they are less toxic to tissues. In addition, by tuning the ratio of the Ln dopants, and excitation with different laser wavelengths, a shift in the color of Ln-based UCNPs was observed. Yao and co-workers have reported a new polymer nanocomposites that consisted of poly-trimethylolpropane trimethacrylate as a polymer matrix and lanthanide orthophosphate: Tb NPs as the non-linear chromophore [545]. By adjusting the ratio of centrosymmetric LaPO₄ to non-centrosymmetric TbPO₄-based hydrate, the LaPO₄:Tb nanomaterials were demonstrated to have SHG properties due to symmetry breaking. These polymer-inorganic nanocomposites have wide spectral tunability from 375 nm to 425 nm for SHG peaks, with an excitation range of 750-850 nm. They also found that the SHG intensity was related to the refractive indices of the materials and was suggested to vary with the frequency of the fundamental wave [546,547]. In the future, controlling the atom number in the formation of the Au nanocluster might offer an alternative solution, in place of Ln-based nanoparticles, for obtaining color-tunable nanoreagents.

Considering that laser operation in photomedicine is required to be user friendly, a single CW laser irradiation that can perform simultaneous PDT/PPT treatments offers a new attractive method to overcome the intrinsic limitations of PDT or PTT alone. PPT alone is of particular benefit to large-area ablation of tumor cancer cells because the heating area is extended. However, the temperature at the periphery of the tumor tissue is less than at the central site and thus unaffected cancer cells can re-grow and migrate to affect healthy tissue nearby. Synchronous PDT treatment could be developed to assist clean-up of the residual cancer cells. On the other hand, the PDT process is oxygen-independent and has limited efficacy in the hypoxia environment. In that situation, the PTT can still do the job well (oxygen unnecessary). Based on this complementary advantage, a new NIR nanoparticle-platform consisting of a Au nanoreagent and a Ce6 photosensitizer was developed, which exhibited single wavelength stimulation of PTT/PDTon cancer cells at 671 nm $(1-2 \text{ W/cm}^2)$ [548]. An ICG-loaded human serum albumin composite (~75 nm), developed by Sheng et al. [174], can simultaneously convert the absorbed light energy to singlet oxygen species and heat upon an excitation at 808 nm laser for 5 min (0.8 W/cm²) thus providing for synergistic PDT/PTT treatment. The combination of PDT and PTT had a higher anticancer efficacy than either single PDT or PTT in an animal study.

In recent years, low-level laser therapy (LLLT) with an NIR laser device has become an increasingly mainstream modality for curing disease [549]. This LLLT system can promote tissue regeneration, reduce inflammation and relieve pain through non-thermal mechanisms [538,550–552]. The most common LLLT management approach includes laser radiation, such as ruby (694 nm), argon (488 and 514 nm), helium-neon (632.8 nm), krypton (521, 530, 568, 647 nm), gallium-aluminum-As (805 or 650 nm), and gallium-AS (904 nm). The proposed LLLT system requires the following [552]:

- (a) A power laser with 0.001–0.1 W of output.
- (b) A wavelength in the range 300–10,600 nm.
- (c) A pulse rate from 0, which is continuous to 5000 Hz (cycles per second).
- (d) Intensity doses of $0.01-10 \text{ W/cm}^2$ and $0.01-100 \text{ J/cm}^2$.

However, for most nanoparticle-mediated tumor depletion, the commonly required power density for photo-thermal therapy is $1-6 \text{ W/cm}^2$, despite the high penetration of NIR light [457,553]. In contrast to high-energy photothermal ablation, the PDT treatment requires a lower power density of incident light. Although this advantage is addressed, most photosensitizers [405] (e.g. phthalocyanine-based derivatives [554,555], porphyrin-based compounds [556,557] and phenothiazine structured dyes [76,558,559]) can only be activated upon exposure to visible wavelengths. To overcome the limitation of the light absorption by tissues, upconversion nanoparticles can convert NIR light into visible light, and the neighboring PS can also be excited through FRET [560]. The ideal design is to overlap the emission spectra of UCNPs in some visible regions with photosensitizer loading. This idea was first presented and explored by Prasad and co-workers [561,562]. Since then, many groups have used UCNPs to convert the deeply penetrating near-infrared light into visible wavelengths for the photo-excitation of photosensitizers and then the production of cytotoxic ${}^{1}O_{2}$, resulting in cancer cell damage via apoptotic and necrotic pathways [458,563].

8. Concluding remarks

The engineered nanoparticles discussed here offer improved, often exotic, physico-chemical properties, and thus nanotechnology will definitely gain by the development of such novel NIR-to-NIR materials. Nevertheless, these developments in nanotechnology poses important questions regarding the impact of these new materials on living systems. Numerous studies have demonstrated the need for nanoparticle risk assessment in relation to a long list of parameters, such as surface charge and (bio)chemistry, size, shape anisotropy, purity, stability, and many others. Such evaluations should contribute to nano-safety considerations during their design and optimization. These existing results also highlight the need to characterize not only NPs, but also the ultimate properties of NPs *in vitro* and *in vivo*, starting from protein adsorption in biological media, colloidal stability, cellular permeability and translocation within cell membranes and whole cells, up to *in vivo* circulation, aggregation, barrier translocation, and finally clearance. This knowledge is important not only for risk assessment of NPs, but also to understand their interaction with biological systems in order to intentionally design enhanced biomedical applications, for use as biomedical imaging, diagnostic tools, and nanoparticle based drug delivery theranostic systems. Moreover, during deep-tissue theranostics, the wavelength and intensity level of NIR excitation should be taken into consideration to achieve the least invasiveness. At excitation wavelengths shorter than 800 nm, endogenous photosensitizers like porphyrins can be excited by two photons, thus generating ROS in tissues. Even though the 1000–1300 nm wavelength range has high penetration in biological tissues, water or pigment absorption induced photo-damage will not be negligible in photoacoustic or high nonlinearity imaging like THG microscopy. A lower pulse excitation rate could help with the relaxation of thermal energy. On the other hand, efficient and multimodal NIR nanomaterials are required to obtain good-enough contrast at low excitation levels.

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